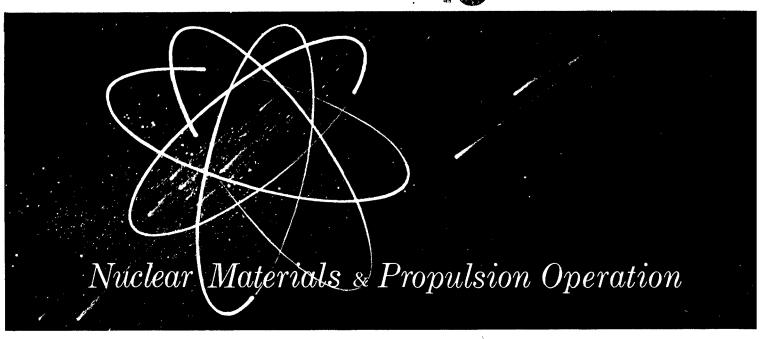
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High-Temperature Materials Program

Progress Report No. 19, Par

JANUARY 25, 1963

GENERAL ELECTRIC

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JANUARY 25, 1963

United States Atomic Energy Commission

Contract No. AT (40-1)-2847

NUCLEAR MATERIALS and PROPULSION OPERATION FLIGHT PROPULSION LABORATORY DEPARTMENT

GENERAL ( ELECTRIC

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### 1. INTRODUCTION AND SUMMARY

### Introduction

This report is the unclassified portion of the nineteenth in a series of monthly reports of the work in process on materials development for the Atomic Energy Commission under Contract AT(40-1)-2847.

Included is a summary of the work from October 15, 1962, to December 15, 1962, on three of the fourteen specific development programs in process. Five of the remaining programs are reported in the classified portion of this report, GEMP-19, Part B. The other six, involving the development of ceramic fuel materials, are reported in alternate months.

### Summary

### HIGH-TEMPERATURE REACTOR MATERIALS FABRICATION RESEARCH (57003)

Stress-rupture results obtained with molybdenum at 2200°C in hydrogen and argon were essentially identical indicating little or no atmosphere effect. Previous data which indicated longer rupture life (and increased creep resistance) in the argon tests have been discounted because of tungsten contamination of the sample material during testing as revealed by chemical analysis.

Identical results at 2200°C in both hydrogen and argon were obtained on two of the three different heats of molybdenum employed in the stress-rupture tests. The third heat exhibited much longer rupture life, particularly at a stress level of 1000 psi. The reason for this difference has not been identified but is probably caused by some material variability.

The stress-rupture properties of tantalum were evaluated in two grades of argon and two grades of hydrogen. The stress-rupture life in argon was found to decrease as the dewpoint was decreased. Tungsten contamination was found on the surface of tantalum samples tested in the "wet" argon due, in all probability, to the transport of tungsten from the furnace heating element by the tungsten-water vapor cycle. This observation stresses the importance of employing extremely low (about -75°C) dewpoints in stress-rupture tests in noble gas atmospheres at elevated (2000°C) temperatures. There was no significant difference noted in the tantalum stress-rupture life test data obtained for tests conducted in commercial (2000 ppm impurities) and high purity grade hydrogen (5 ppm impurities).

### EFFECTS OF RADIATION ON HIGH-TEMPERATURE ALLOYS (57004)

The preliminary experimental program on the refractory metals was initiated with the irradiation of several tungsten and molybdenum flat test specimens. Support equipment and furnaces for testing at high temperature in hydrogen or inert atmosphere or in vacuum are being assembled.

All the tensile data generated on A-286, Hastelloy X, and Rene' 41 has now been reported. The effects of irradiation on the stress rupture properties of overaged and underaged specimens of Rene' 41 are presented.

Preliminary studies on the creep-rupture properties of Hastelloy X are illustrated in the form of specimen elongations as a function of testing time for various temperatures and stresses.

Studies on the number of atom displacements in iron due to neutrons of various energies are discussed. The angular distribution of neutron scattering in iron and also the effect of the angle of neutron incidence on the medium are presented. The influence of material geometry on the number of atom displacements per unit flux is demonstrated by comparing the displacements in an iron column with those in an iron slab for various base widths and thicknesses.

# LONG TERM IRRADIATION TEST OF 80Ni - 20Cr FUEL ELEMENT SPECIMEN (57301)

No further work was done during this reporting period. A final summary will be included in the 1963 High-Temperature Materials Annual Report.

# 2. HIGH-TEMPERATURE REACTOR MATERIALS FABRICATION RESEARCH

(57003)

The purpose of this program is to develop and evaluate methods of preparing and joining refractory metals and alloys, and other high-temperature materials for use as fueled and non-fueled high temperature ( $1000^{\circ}$  to  $3000^{\circ}$ C) reactor components.

#### STRESS-RUPTURE TEST RESULTS

#### Molybdenum

Stress-rupture testing of molybdenum at 2200°C in hydrogen and argon was continued in this reporting period. Previous stress-rupture tests\*† were performed by loading the specimen before heating. During this report period, equipment was modified to allow the load to be applied after the specimen reached test temperature. With this technique, an unstressed heat treatment of the specimen can be obtained and the opportunity to obtain first-stage creep data is enhanced. No significant variation from previously reported stress-rupture data has been observed at the temperatures used in tests to date.

The tests performed during this reporting period on the three heats of molybdenum, designated Mo-A, Mo-B, Mo-C, are summarized in Table 2.1. All of the Mo-A, Mo-B, Mo-C stress-rupture results obtained to date are shown in Figures 2.1 and 2.2. These data show that Mo-B (0.011 inch thick) and Mo-C (0.015 inch thick) have equivalent stress-rupture properties at 2200°C in both hydrogen and argon at the stress levels investigated. Mo-A (0.010 inch thick), in argon and hydrogen, exhibits a rupture life, at equivalent loads, which is much longer than that of either Mo-B or Mo-C, particularly at stress levels above 400 psi. However, if the ten-hour stress-rupture levels are extrapolated to 100-hour life, the variation between Mo-A and Mo-B or Mo-C appears to be negligible. The results of chemical analyses of as-received Mo-A, Mo-B, and Mo-C are so similar that strength differences cannot readily be attributed to variations in material composition. Additional studies involving electron microscopy and electron diffraction evaluations are now under way to investigate possible intergranular deposits which might be responsible for these strength differences.

Photomicrographs of "as-received" material are shown in Figure 2.3. While these structures appear identical, a more detailed examination at higher magnification revealed incipient recrystallization in the Mo-A sample but none in Mo-B. Hardness measurements for as-received material were 252 DPH for Mo-A and 275 DPH for Mo-B.

<sup>\*&</sup>quot;High-Temperature Materials Program Progress Report No. 17, Part A," GE-NMPO, GEMP-17A, November 15, 1962,

p. 19.

†"High-Temperature Materials Program Progress Report No. 15, Part A," GE-NMPO, GEMP-15A, September 14, 1962, p. 21.

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0.1



TABLE 2.1

HIGH-TEMPERATURE STRENGTH D	ATA FOR	MOLYBDENUM
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								`	•	
			ρ <sup>‡</sup> <b>HIGH-TEM</b> I		TABLE	2.1				
		ζ,	HIGH-TEM	PERATURE ST	RENGT	I DAT	A FOF	NOL	YBDENUI	М
			i	·	· · · · · · · · · · · · · · · · · · ·		То Е		ndicated	Percent Elongation
		,	remperature,		Stress,	Filor	ngation		<u> </u>	In 1-Inch
	Material		°C	Atmosphere	psi	2	5	10	Rupture	Gage Length
	Mo-A	_	2200	Н2	550	0.4	1.2	2.65	6.70	29.7
	Mo-A		2200	H <sub>2</sub>	600	0.2	0.8	2. Ò	6.18	30.9
	Mo-A		2200	H <sub>2</sub>	700	0.2	0.8	1.6	3.44	26.0
	Mo-A		o 2200	H <sub>2</sub>	800	0.2	0.7	1.4	2.83	20.2
	Mo-A		2200	H <sub>2</sub>	1000	0.2	0.5	1.0	1.69	20.9
	Mo-A		s / <b>2200</b>	A	600	0.5	1.5	3.0	5.75	14.3
	Mo-A		· 2200	A	700	0.4	1.0	2.2	4.30	21.3
	Mo-A	,	2200	A	900	0.2	0.7	1.6	2.75	19.2
	Mo-A		2200	Ā	1100	0.1	0.4	1.0	1.56	17.5
	Mo-A	,	2200	A	1300	0.2	0.4	0.8	1.10	16.0
	Мо-В		2200	$_{ m H_2}$	400	0.3	1.0	2.0	6.7	50.4
	Mo-B	,	· 2200	$_{ m H_2}^{ m 2}$	500	0.1	0.4	1.1	3.22	47.5
	Mo-B	€	2200	$H_2$	600	0.1	0.3	0.6	1.49	49.5
	Mo-B		2200	$\mathtt{H}_{2}^{-}$	700	0.06	0.2	0.4	0.81	58.1
	Mo-B		2200	A	400	0.3	0.9	2.0	5.57	37.2
	Mo-B	C.	2200	A	500	0.1	0.4	1.0	2.95	59.4
	Mo-B	c	2200	A	600	0.1	0.3	0.5	1.33	54.2
	Mo-B	,.	· ^ <b>22</b> 00	A	700	0.06	0.1	0.3	0.58	30.4
•	Mo-C		2200	H <sub>2</sub>	300	1.1	2.8	5.2	12.05	60. <b>2</b>
	Mo-C	$\circ$	2200	$_{ m H_2}$	450	0.2	0.7	1.3	2.72	64.0
	Mo-C		2200	H <sub>2</sub>	550	0.08		0.5	1.48	90.5
	Mo-C		2200	$H_2$	600	0.07		0.4	1.28	105.5
	Mo-C	. '	2200	$_{ m H_2}$	750	0.04		0.2	0.53	<b>57.2</b>
	Mo-C	e.	2400	$H_2$	400	0.03	0.08	0.2	0.4	61.5
	Mo-C	٠,٠	2200	A	300	1.3	2.9	5.2	15.37	70.6
	Mo-C		2200	A	450	0.2	0.7	1.4	3.25	47.2
	Mo-C		2200	A	550	0.1	0.3	0.7	1.87	103.8
	Mo-C		2200	A	700	0.08	0.1	0.2	0.45	78.8
	Mo-C		2200	A	900	0.01		0.07	0.16	79.7
	Mo-C		2400	A	400	0.03	0.07	0.2	0.31	66.0

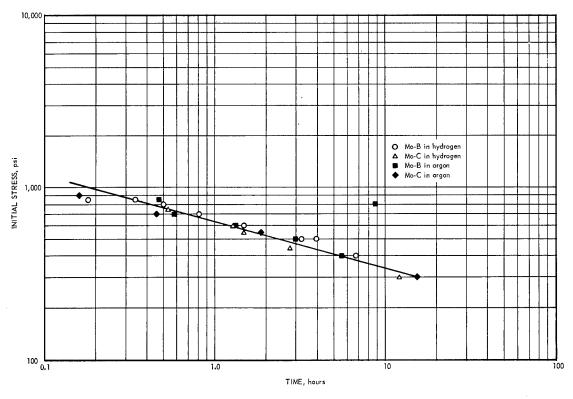


Fig. 2.1 — Results of Mo-B and Mo-C stress-rupture tests in hydrogen and argon at 2200°C

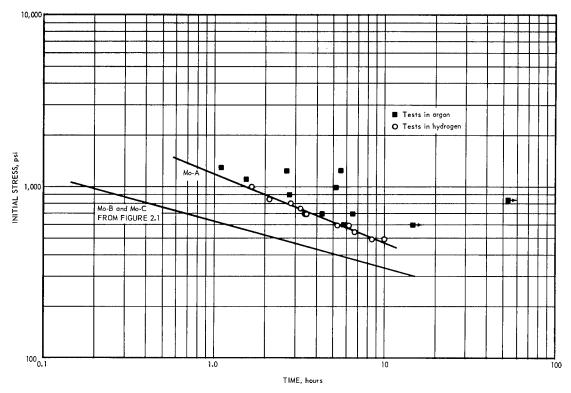


Fig. 2.2-Results of Mo-A stress-rupture tests in hydrogen and argon at  $2200^{\circ}$ C

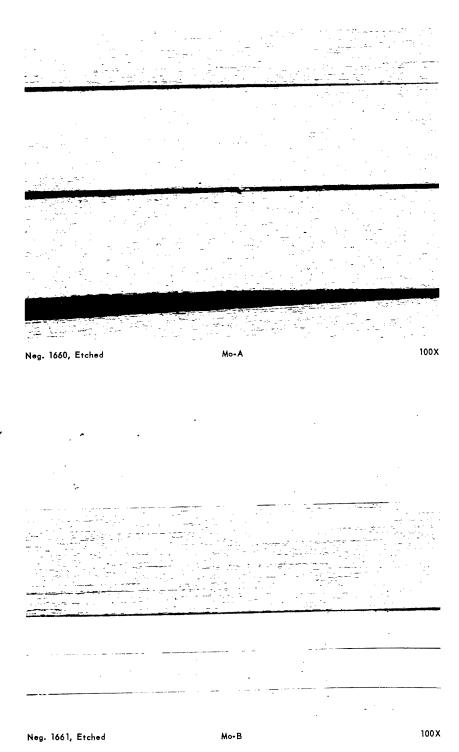


Fig. 2.3 — Photomicrographs of as-received Mo-A and Mo-B  $_{\rm material}$ 

Typical microstructures of Mo-A and Mo-B after testing at 2200°C and 500 psi in hydrogen are shown in Figure 2.4. The Mo-A specimen which ruptured after 10 hours at temperature exhibited small uniform and equiaxed grains with grain boundary separation increasing near the fracture. The Mo-B specimen which ruptured after 3.9 hours at temperature exhibited very large and equiaxed grains. This difference in grain size was noticed consistently and may in some way be responsible for the observed difference in strength. Another difference noted involves a higher recrystallization temperature for Mo-B.

Figures 2.5, 2.6, and 2.7 show typical specimens of Mo-A, Mo-B, and Mo-C after test and point out the differences in types of fracture at failure as well as a comparison of the elongations.

Some previous tests of molybdenum in argon produced results which were markedly different from those obtained in hydrogen at the same temperature and stress level. Not only was the rupture life higher, as shown in Figures 2.1 and 2.2, but the creep resistance was observed to be increased by an order of magnitude. Chemical analyses have since revealed that these specimens were contaminated with tungsten which probably accounts for the observed increase in strength. The source of this tungsten is obviously the furnace heating element. Since vapor pressures are too low to account for such transport and no tungsten contamination was found in other specimens heated in the high purity gases normally used to conduct these tests, the contamination must be due to the formation of the volatile tungsten oxide resulting from either the tungsten-water vapor cycle or the reaction of tungsten with air which entered the furnace through defective seals. The fact that this unusual strength is observed only occasionally points to some inadvertent deviation from ordinarily carefully controlled test conditions. Some proof of the contaminating mechanism was obtained when a molybdenum specimen was deliberately coated with a thin film of tungsten oxide and tested in argon. This test showed exactly the same increased strength referred to above. Interestingly, this same test in hydrogen failed to show any property increase. Since in this test the film of tungsten oxide was quickly reduced to a film of tungsten, it seems that the strengthening mechanism must be associated with the presence of a volatile oxide phase. Further tests are planned to more clearly define this phenomenon.

Chemical analysis of three of these Mo(W) specimens showed that the amount of tungsten absorbed was a function of the time the test sample was at temperature. An electron microprobe analysis of one of these samples showed that the concentration of tungsten was high (22%) on the surface and low (1 to 2%) in the center of the material. No tungsten contamination was observed in specimens which exhibited a stress-rupture life falling close to the lines in Figures 2.1 and 2.2. Obviously, the test results in argon are affected only when the furnace atmosphere is inadvertently contaminated with tungsten oxide.

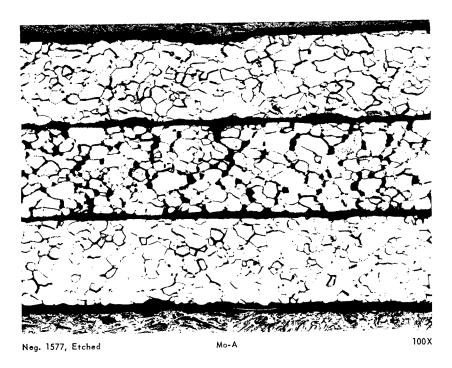
#### Tantalum

Previous stress-rupture tests of tantalum at 2600°C in argon have shown a certain amount of scatter compared to the fairly reproducible results obtained in the hydrogen tests. In order to determine the effect of various atmospheres on identical tantalum specimens, a series of tests was conducted using two sources of hydrogen and two sources of argon as defined below:

- 1. Hydrogen from the plant distribution system
- 2. Bottled high purity hydrogen
- 3. Argon from the plant distribution system
- 4. Bottled liquid argon

Table 2.2 presents a summary of these 2600°C tests with a specimen stress of 170 psi.

14



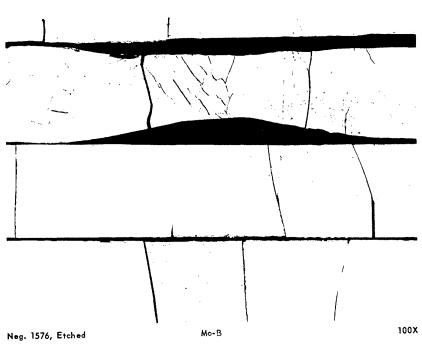


Fig. 2.4 — Photomicrographs of Mo-A and Mo-B tested at 2200°C and 500 psi in  $\rm H_2$  for 10 and 3.9 hours, respectively

No. 1	No. 2	No. 3	No. 4	No. 5	No. 6	No. 7
						O
,		Access 1970				
		m (3) 3	m <sub>Go</sub> A			
600 -	500	800	TIAL STRESS, p	1300	1100	900
		TIME	TO RUPTURE,	hours		
6.18	6.7	2.83	1.69	1.10	1.56	2.75
	PE	ERCENT ELONGA	ATION IN 1-INCH	H GAUGE LENGT	TH.	i
30.9	29.7	20.2	20.9	16.0	17.5	19.2
		TE	EST ATMOSPHER	RE		
Hydrogen	Hydrogen	Hydrogen	Hydrogen	— Argon	Argon	Argon
			DEW POINT, °C			
<b>-55</b>	55	<b>-55</b>	<b>–</b> 55	<u>-67</u>	<b>–67</b>	<b>-72</b>
					1 000000	

Fig. 2.5-0.010-inch-thick Mo-A specimens stress-rupture tested at 2200°C (Neg. P62-11-4)

The data summary in Table 2.2 indicates an increase in the stress-rupture life of tantalum as the dewpoint of the argon is increased. This is attributed to a more active tungstenwater vapor reaction at the higher moisture levels which results in larger amounts of tungsten or tungsten oxide being transported to the specimen, thereby yielding increased strength. This effect on stress-rupture properties is quite similar although not as pronounced as that observed with molybdenum tested in argon. Chemical analyses are now in progress to evaluate these anticipated differences in contamination. If this mechanism is confirmed, the scatter obtained in the argon tests can be attributed to dewpoint variations between tests.

The tantalum results obtained in the two different hydrogen atmospheres were essentially identical. Obviously, differences in the impurity levels of the hydrogen exhibit a very minimal effect on this material under these conditions.

The above considerations emphasize the importance of employing noble gases of extremely low (-75°C) dewpoints for use in high-temperature testing of refractory metals.

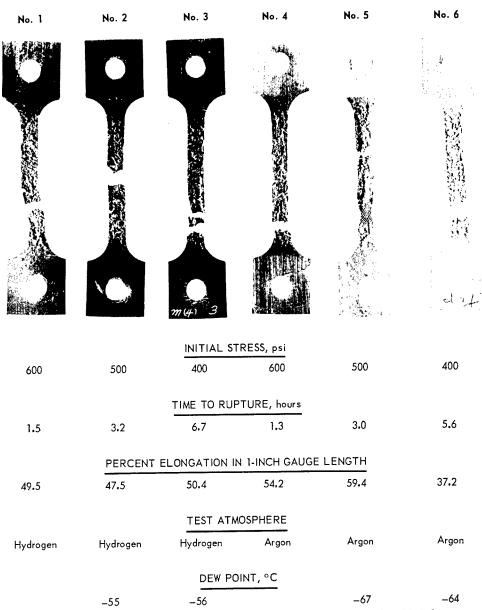


Fig. 2.6-0.011-inch-thick Mo-B specimens stress-rupture tested at 2200°C (Neg. P62-11-8A)

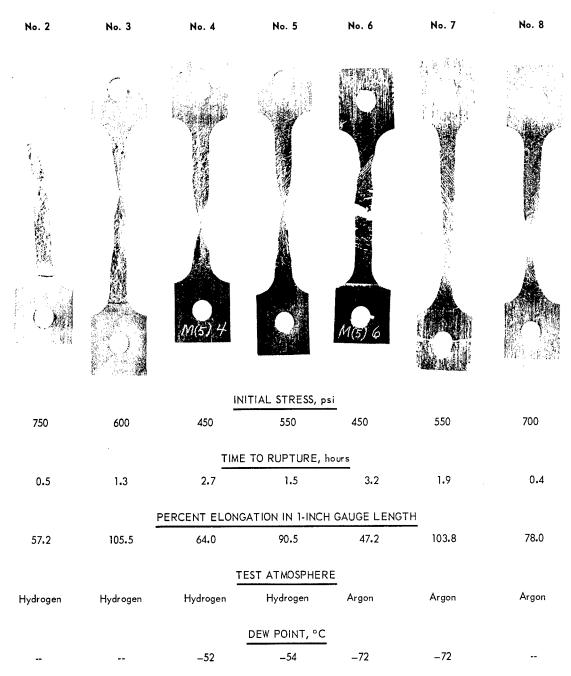


Fig. 2.7 -0.015-inch-thick Mo-C specimens stress-rupture tested at 2200°C (Neg. P62-11-8B)

TABLE 2.2

EFFECT OF ATMOSPHERE ON TANTALUM STRESSRUPTURE TESTS AT 2600°C AND 170 PSI

Atmosphere	Dewpoint, OC	Rupture Time, hr	Percent Elongation In 1-Inch Gage Length
Argon <sup>a</sup>	-59	5.03	45
Argona	-65	3.83	58
Argonb	<b>-</b> 79	2.33	47
Argonb	-79	2.81	40
Hydrogena	-54	2.05	40
Hydrogen <sup>a</sup>	-53	2.06	36
Hydrogen <sup>c</sup>	-79	2.67	57
Hydrogenc	-79	2.23	30

aplant distribution system.

Hydrogen is shown to be an attractive test atmosphere for such measurements, particularly since the tungsten-water vapor cycle is almost completely suppressed by the hydrogen.

#### Analysis of Creep Measurements

Further analysis of the total elongation measurements obtained during the stress-rupture tests is being made in order to obtain a preliminary evaluation of creep behavior. As previously reported\* a plot of log initial stress versus log linear creep rate for Ta - 10W at 2600°C and 2800°C was linear with a slope of approximately 0.25 yielding an exponent on stress in the Weertman† equation equal to about 4.0. This was pointed out as being very close to the stress relationship developed by Weertman for dislocation climb. Data for several additional materials are presented in Figure 2.8 and are seen to follow the same relationship, i.e., a slope of 0.25. An exception is noted in the case of Mo-A which has a slope close to 0.6. While the exact cause of this different creep-dependence on stress is unidentified, this difference is reflected in the stress-rupture data. Creep analyses‡ have shown that the slope of the lines in Figure 2.8 should be equal to that of the corresponding line in Figure 2.1 except of opposite sign. Therefore the different slopes obtained in Figure 2.1 for different molybdenum materials should be reflected as different slopes in the creep rate plot of Figure 2.8. These data provide a measure of the reproducibility and accuracy of the testing in this program.

#### WORK PLANNED FOR NEXT PERIOD

Stress-rupture testing at 2200°, 2400°, 2600°, and 2800°C will be continued on refractory alloys in hydrogen and argon, and initiated in neon.

Creep measurements within the specimen gage section will be made on a selective basis to confirm the creep analyses mentioned above based on elongation measurements.

Longer time (100 hour) stress-rupture tests will be performed to determine the validity of extrapolating current 10-hour data.

<sup>&</sup>lt;sup>b</sup>Bottled liquid argon.

<sup>&</sup>lt;sup>c</sup>Bottled high purity hydrogen.

<sup>\*&</sup>quot;High-Temperature Materials Program Progress Report No. 17, Part A," GE-NMPO, GEMP-17A, November 15, 1962,

<sup>&</sup>lt;sup>†</sup>J. Weertman, Journal of Applied Physics, Volume 26, 1955, p. 1213.

<sup>&</sup>lt;sup>‡</sup>J. B. Conway, "Presentation, Correlation, and Interpretation of High-Temperature Stress Rupture and Creep Data," GE-NMPO, TM 62-11-14, November 1962.

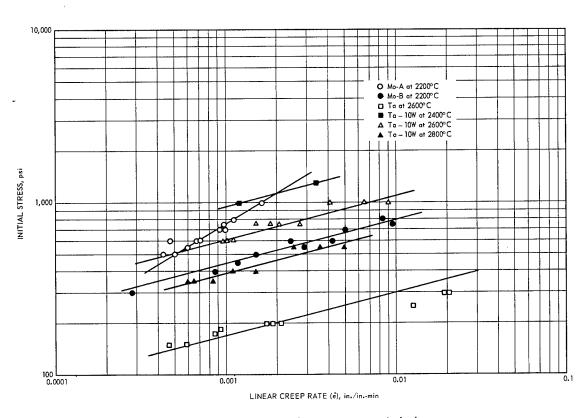


Fig. 2.8 - Initial stress versus linear creep rate in hydrogen

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# 3. EFFECT OF RADIATION ON HIGH-TEMPERATURE ALLOYS

(57004)

The objective of this program is to determine the effect of radiation on the time-, temperature-, and stress-dependent properties of selected high-temperature alloys and refractory metals, to identify the causes of any observed changes in these properties, and to develop remedial measures.

#### REFRACTORY METALS EXPERIMENTAL PROGRAM

A program to study the influence of neutron irradiation on refractory metals and their alloys has been initiated. This program will follow the eleven high-temperature alloy experiments based on A-286, Hastelloy X, and Rene' 41 materials. The refractory metals program will be the major experimental effort for Calendar Year 1963 with only a few concluding experiments to be performed on the high-temperature alloys.

The objective of the refractory metals program is to determine the relative effects of irradiation on the physical and mechanical properties of these materials. Primary emphasis will be placed on establishing damage mechanisms and remedial measures. The major effort will be conducted on tungsten and tungsten-base alloys with a limited effort, primarily for background information, on molybdenum, molybdenum-base alloys, and other body-centered metals.

The physical and mechanical tests will include creep-rupture, tensile, brittle-ductile transition temperature, hardness, and resistivity. Other tests may be included, as deemed necessary, to help establish damage mechanisms. Emphasis will be placed on the influence of grain size in radiation effects. An investigation of the effect of various post-irradiation heat treatments on the properties of refractory metals is planned. Both isochronel and isothermal annealing studies will be performed. The use of materials of several purity levels, including one of extremely high purity, is also being considered. Initial creep-rupture tests will be conducted at approximately 1650°C. Tensile testing will be performed at room and at elevated temperatures.

An initial exploratory study of the creep-rupture properties of refractory metals is in progress. Molybdenum and tungsten specimens were irradiated at reactor ambient temperatures to a neutron dosage of approximately 3 x  $10^{17}$  and 3 x  $10^{18}$  nvt ( $E_n \ge 1$  Mev), respectively. The predicted induced activities of some refractory metals are shown in Figure 3.1. The experimental data for tungsten and molybdenum will be compared at a later time. For the present, the experimental program on tantalum and tantalum-base alloys will be kept to a minimum due to the long residual induced radioactivity.

Two creep-rupture stands used in the FY-62 program are being modified for the refractory metal program. A furnace designed for tensile or stress-rupture testing up to 1780°C will be installed. The furnace will be equipped to test specimens in hydrogen,

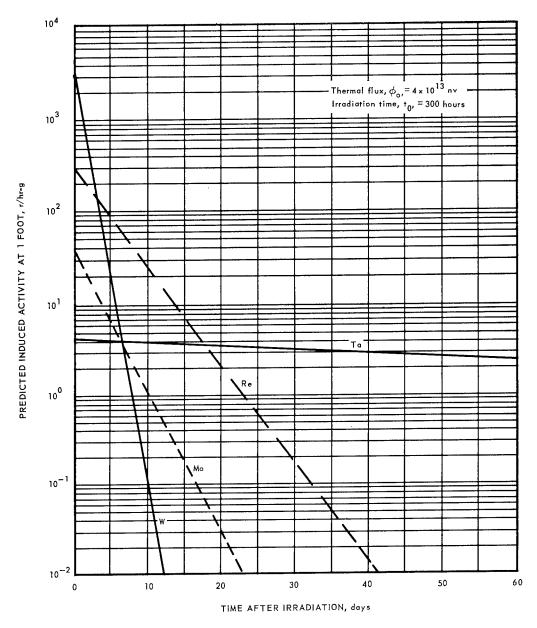


Fig. 3.1-Induced activity of refractory metals as a function of decay time

inert gas, or in vacuum. It will employ a circular molybdenum heating element. Tungsten heating elements will be prepared for later installation. It is planned to have this furnace in operation early in the next report period.

#### Specimen Preparation

Commercially pure 0.135- and 0.200-inch diameter tungsten rods were received for the preliminary investigation. Metallographic examinations and density measurements were completed; complete chemical analyses are in progress. Three 40-inch lengths of 0.135-inch-diameter W - 25Re rod were also received and are being evaluated. The preliminary designs of the specimens for the refractory metals program are shown in Figure 3.2. The flat specimens will be used in the initial work since extensive control data exists for the refractory metals based on this specimen type. Based on the comparison of the round and the flat irradiated specimens data, one design will be selected for the major

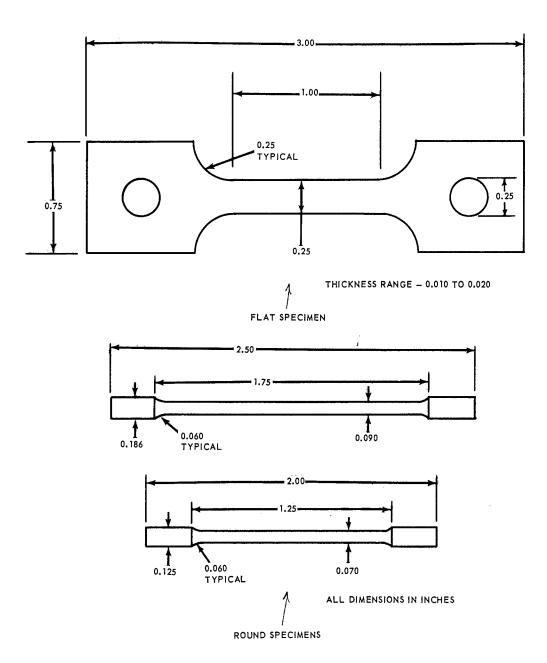


Fig. 3.2-Preliminary design of refractory metal specimens for creep, stress-rupture, and tensile testing

testing program on refractory metals. Manufacture of specimens and load train hardware from tungsten is in progress and will be completed within the next reporting period.

The surface condition of tungsten specimens is known to affect its mechanical properties.\*†
A limited exploratory study was conducted to investigate the techniques required for electropolishing tungsten. Equipment, as shown in Figure 3.3, was assembled and installed. This apparatus permits both rotating the specimen and stirring the electrolyte. Polishing conditions, similar to those of another investigation,‡ include the following:

- 1. Electrolyte, 2 percent sodium hydroxide
- 2. Current density, 3 to 6 amp/sq. in.
- 3. Rotation of specimen, 3 rpm

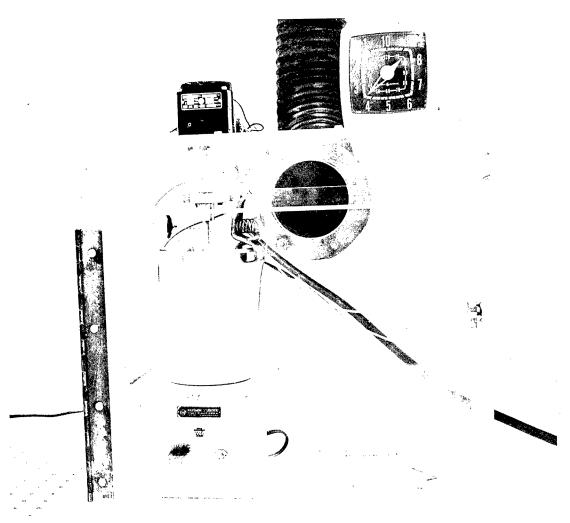


Fig. 3.3 - Refractory metal specimen surface electropolishing apparatus (Neg. P62-12-5B)

<sup>\*</sup>K. Sedlatschek, and D. A. Thomas, "The Effect of Surface Treatment on the Mechanical Properties of Tungsten," Powder Metallurgy Bulletin, Vol. 8, Nos. 1-2, June 1957, pp. 35-40.

<sup>&</sup>lt;sup>†</sup>J. R. Stephan, "Effect of Surface Condition on Ductile-To-Brittle Transition Temperature of Tungsten," NASA TN D-676, Lewis Research Center, February 1961.

<sup>&</sup>lt;sup>‡</sup>J. R. Stephan, "An Exploratory Investigation of Some Factors Influencing The Room-Temperature Ductility of Tungsten," NASA TN D-304, Lewis Research Center, August 1960.

- 4. Cathode, stainless steel beaker (1.5 qt., 4 inch I.D.) centered about the rotating specimen
- 5. Temperature of electrolyte, 25°C

The bottom of the beaker is fitted with a Teflon disc to prevent reaction and gas evolution.

Limited experimentation with tungsten specimens have demonstrated that it is possible to electropolish the surface from approximately a 24 microinch RMS ground finish to a highly polished surface of 2 to 5 microinch RMS finish. Rates of metal removal are relatively rapid, approximately 0.5 to 1.0 mil from the gage diameter per minute. Figure 3.4 shows the finish of machined tungsten before and after electropolishing.

#### HIGH-TEMPERATURE ALLOYS EXPERIMENTAL PROGRAM

#### Status of Irradiation Program

The status of the experimental program is summarized in Table 3.1.

Test capsule MT-99, containing smooth stress-rupture specimens of A-286 alloy, was prepared and shipped to GE-ITS for irradiation. These specimens will be used (Experiment No. 6) for the determination of the effect of neutron reactions with the trace quantities of boron present in this alloy. There are four groups of seven specimens each. The first three groups, from the same heat, contain 0.00085, 0.00425, and 0.010 weight percent natural boron. The fourth group of seven specimens contains 0.0065 weight percent boron which is depleted in the boron-10 isotope. The test capsule will be inserted in the K-3 hole of the ETR and will be irradiated at reactor ambient temperatures for two cycles. The capsule is instrumented with nickel-cobalt flux monitors attached to center rods.

Capsule LTTS-18 irradiations were completed in the Graphite Pile facility at Oak Ridge. This capsule contains the A-286 test specimens that will be used in the relaxation tests (Experiment No. 9). The capsule will be removed from the reactor the first week of the next reporting period and the specimens shipped to Evendale. Testing of control specimens will be initiated following evaluation of the irradiation temperature data. The capsule was irradiated approximately 490 hours at a temperature greater than 620°C. The temperature gradient along the specimen length in stage 1 was approximately 80°C and 30°C in stage 2. The nominal temperature difference between the two stages was approximately 45°C.

The irradiation plans for Rene' 41 relaxation specimens have not been established pending evaluation of the A-286 specimens. In order to place greater effort on the new refractory metals program, the Rene' 41 portion of experiment 9 is being postponed indefinitely.

#### Post-Irradiation Testing

Stress-Rupture Results for Experiment No. 4 - Experiment No. 4 was conducted to evaluate the effects of irradiation on the precipitation hardening alloys A-286 and Rene' 41 in the underaged and the overaged heat treated condition. The post-irradiation test results for the A-286 alloy were reported previously.\*

The Rene' 41 test data are presented in Table 3.2 and Figure 3.5. These specimens were irradiated in Capsule MT-38 at approximately 650°C. The overaged control data indicates higher rupture strengths than those of the underaged specimens. A similar effect was noted for the irradiated specimens. Both the control and the irradiated overaged and underaged test data indicate lower rupture strengths than the data obtained from test specimens† given the standard heat treatment. It is interesting to note that the underaged control

<sup>\*&</sup>quot;High-Temperature Materials Program Progress Report No. 9, Part A," GE-NMPO, GEMP-9A, March 30, 1962, p. 44.

<sup>†&</sup>quot;High-Temperature Materials Program Progress Report No. 11, Part A," GE-NMPO, GEMP-11A, May 15, 1962, p. 53.

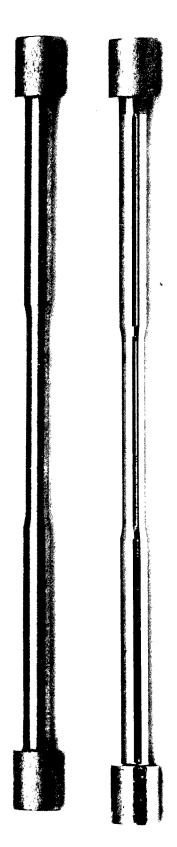


Fig. 3.4 - Polished and unpolished tungsten specimen (Neg. P62-12-4)

TABLE 3.1

				TABLE 3.1			11 -	15	عنون بها		
			SUMMARY	OF EXPERIMEN	TAL PROGRAM		$\Delta^{!}$				
				Irradiation	Conditions						
				Approximate	Approximate			Rup	ture Strengt	h, 1000 p	91
periment				Temperature,	Dosage,		Temperature, °C	100 H Control	ours Irradiated		Hours <sup>a</sup>
No.	Objective	Alloys	Capsule	°C	nvt (E <sub>n</sub> ≥ 1 Mev)	Heat					
1	To investigate the effects of neutron	A-286	MT-27	540	7 x 10 <sup>19</sup>	C2527	540	100	93	82	77
•	irradiation on the tensile and stress		MT-38	650	5 x 10 <sup>19</sup>	C2527	650	54	38	41	22
	rupture properties of high-temperature		MT-43	650	11 x 10 <sup>19</sup>	B15649	650	52	37	35	19
	alloys receiving the standard heat treat-		MT-43	650	11 x 10 <sup>19</sup>	7611 &	650	49	34	31	18
	ment for these experiments. To generate					12878					
	the background data needed on these	Hastelloy X	MT-27	540	$7 \times 10^{19}$	E9500	540	66	55	54	44
	alloys for comparison with other	11101011119	MT-38	650	5 x 10 <sup>19</sup>	E9500	650	43	37	27	27
	experiments		MT-43	650	11 x 10 <sup>19</sup>	X1-4385	650	43	42	30	32
			MT-51	870	4 x 1019	E9500	870	9	8	4	5. 5
			MT-51	870	4 x 10 <sup>19</sup>	E9500	650	41	-	27	-
		Rene' 41	MT-38	650	5 x 10 <sup>19</sup>	TV746	650	107	85	87	67
		( Rene 41)	MT-43	650	11 x 10 <sup>19</sup>	TV754	650	124	101	101	77
			MT-51	870	4 x 10 <sup>19</sup>	TV754	870	23	18	12	13
			MT-51	870	4 x 10 <sup>19</sup>	TV754	650	-	104	-	92
			M1-31								65
2	To investigate the effect on the stress	A-286	MT-27	540	7 x 10 <sup>19</sup>	C2527	540	97	86	76	
	rupture properties of a standard heat		MT-38	650	5 x 10 <sup>19</sup>	C2527	650	54	31	40	17
	treatment following neutron irradiation.	Hastelloy X	MT-27	540	7 x 10 <sup>19</sup>	E9500	540	61	66b	52	57b
	**	nasterioy X	MT-38	650	5 x 1019	E9500	650	39	35	25	23
			MT-51	870	7 x 10 <sup>19</sup>	E9500	870	9.5	6. 5	5. 5	3
		727.5							92	93	87
		(Rene' 41)	MT-38	650	5 x 10 <sup>19</sup>	TV746	650	105			4
		<u>.</u>	MT-51	870	4 x 1019	TV754	870	26	11	12	•
3	To investigate the role of grain size	A-286	MT-43	650	11 x 1019	B15649	650	57	42	43	28
3	and other related structural charac-			***			650	120	91	96	66
	teristics in the possible mechanisms	(Rene' 41 )	MT-43	650	11 x 10 <sup>19</sup>	TV754			11b	13 <sup>b</sup>	8b
	of neutron-induced stress rupture		MT-51	870	4 x 10 <sup>19</sup>	TV754	870	20b	11-	130	6-
	property changes.										
	property changes.				10		540	96	80	74	70
4	To investigate the effect of the degree	A-286	MT-27	540	7 x 1019	7611 7611	650	52	41	40	28
	of pre-irradiation aging on the stress		MT-38	650	. 5 x 10 <sup>19</sup>	7611	650				
	rupture properties of irradiated alloys.	Rene' 41	MT-38	650	5 x 10 <sup>19</sup>	TV746	650	Testir	g completed	l. See Fig	ures 3.5 and 3
		trene 11	MT-51	870	4 x 10 <sup>19</sup>	TV746	870	-	-	-	-
							650	57	54	38	35
5	To determine whether some parameter	A-286	MT-35	36	3.0 x 10 <sup>16</sup>	C2527	650	57	54	38	35
	that is a function of thermal neutrons				12.0 x 10 <sup>16c</sup>	C2527	630	31	**	-	•••
	has an effect on stress rupture propertie	s.									
				••	3.0 x 10 <sup>16</sup>	C2527	650	57	46	38	30
			MT-36	36	12. 0 x 10 <sup>16</sup>	C2527	650	57	46	38	30
						C2321					
6	To investigate the effect of neutron re-	A-286	MT-99	Ambient	7 x 1020	-	650	Shippe	to ITS. Sci	heduled fo	r irradiation
•	actions with trace quantities of boron							about a	January 7, 1	963.	
	generally present in alloys on their										
	stress rupture properties.										
_			1 mmc 15	36	3 x 10 <sup>19</sup>	C2527	25	Irradia	tion comple	ted. Test	ing in progress
7	To determine whether neutron irradiatio		LTTS-15					Contro	l specimens	testing a	pprox. 90%
	at reactor ambient temperature induces	Hastelloy X	LTTS-15	36	3 x 10 <sup>19</sup>	E9500	25	comple			
	strain rate embrittlement.	Rene' 41	LTTS-15	36	3 x 10 <sup>19</sup>	TV746	25				
									50d	35	32d
8	To determine the fast neutron threshold	A-286	LTTS-17	650	8. 3 x 10 <sup>14</sup>	B15649	650	52 52	50u 42d	35 35	25d
	of radiation-induced changes for stress		LTTS-16	650	2.7 x 1016	B15649	650	52	424	30	
	rupture properties of high temperature	Hastelloy X	LTTS-17	650	8.3 x 10 <sup>14</sup>	E9500	650	43	42	27	27
	alloys.	nasterioy A	LTTS-16	650	2. 7 x 10 <sup>16</sup>	E9500	650	43	37	27	27
							050	124	124d	101	101d
		Rene' 41	LTTS-17	650	8. 3 x 10 <sup>14</sup>	TV754	650	124	113d	101	90d
			LTTS-16	650	2.7 x 10 <sup>16</sup>	TV754	650				
9	To investigate the effect of neutron	A-286	LTTS-18	650	1019	B15649	-	Irradia	ation comple	te. To be	shipped to
9		A-200	21.10 10					Evenda	ıle.		
	irradiation on the relaxation properties										
	of high-temperature alloys.										
10	To investigate the effect of neutron	A-286	MT-51	540	$4 \times 10^{19}$	B15649	540	-	-		-
10	irradiation on the high-temperature	A-200	MT-51	650	4 x 10 <sup>19</sup>	B15649	650	69e	-	56 <sup>f</sup>	-
	fatigue properties of high-temperature						650	_ •	_	_	_
		Rene' 41	MT-51	650	4 x 10 <sup>19</sup>	TV746	650	•	-	-	-
	alloys.						870	8	7. 5	3. 5	-
		Hastellov Y	MT-51	870	4 x 10 <sup>19</sup>	E9500					
11	To investigate the effect of neutron	Hastelloy X			4 x 10 <sup>19</sup> 4 x 10 <sup>19</sup>	E9500	650	44	35	32	25
11		Hastelloy X	MT-51 LTTS-14 MT-51	870 650 870							25 5, 6

RExtrapolation data.

DTentative data based on partially completed experiment.

CThermal flux.

dApproximate.

Stress value listed is the endurance limit for 10<sup>7</sup> cycles.

Istress value listed is endurance limit for 10<sup>8</sup> cycles.

TABLE 3. 2
STRESS RUPTURE PROPERTIES OF RENE' 41 AT 650°C

		, , ,	EDD ROT I	EXPERIMEN	T NO. 4		
" 1 ( ).	()	Specimen <sup>a</sup>	Stress, psi	Rupture Time, hr	Elongation, % in 1.5 in.	Reduction Of Area,	
O		Controls - Si	mulated cy	cle, 100 hours at	650°C:		
04	o'	Overagedb 29R c' 30R	100, 000 95, 000	140. 5 178. 4	2. 8 6. 9	4.9 7.8	, "
06	o`	Underaged <sup>C</sup> 19R · · · 20R · · ·	100, 000 95, 000	30.6 29.0	1. 3 1. 1	0. 8 0. 8	
		Irradiated at	650°C in N	MT-38: Var Ca	doed cons	erzoer od cope	31
O.A.	o	Overagedb 26R 28R	90, 000 82, 000	45. 6 110. 0	1. 6 2. 0	3. 2 6. 3	
<u>, \$</u>		Underaged <sup>C</sup> 16R 17R 18R	90, 000 85, 000 70, 000	3. 0 1. 5 7. 4	1. 7 0. 9 1. 6	6. 3 3. 2 nil	

aAll specimens were of the smooth-bar type from Heat TV-746.

bHeat treatment prior to grinding:

1205°C - 2 hr - water quench

 $1065^{\circ}C$  - 6 hr - water quench

 $925^{\circ}C$  -  $48 \; hr$  -  $air \; cool$ 

760°C - 100 hr - air cool

 $^{\rm C}{\rm Heat}$  treatment prior to grinding was the same as the underaged specimens plus 1175°C for 1/2 hr and cooled to 25°C in argon.

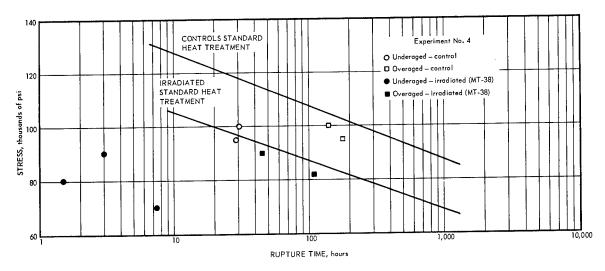


Fig. 3.5 - Stress-rupture properties of Rene'41 at 650°C - Heat TV-746, Capsule MT-38

specimen data appear to have rupture strengths comparable to the irradiated specimens which received the standard heat treatment prior to irradiation.

The Rene' 41 data for specimens irradiated at 870°C in Capsule MT-51 and post-irradiation tested at the same temperature are listed in Table 3. 3 and plotted in Figure 3. 6. The control test data indicated no significant differences, based on the few specimens tested, between the overaged and the underaged heat treatment, and, therefore, the data were grouped in the curve fitting. As reported in the 650°C tests, the irradiated overaged specimen test data show greater rupture strengths than the underaged specimen data. The data seem to indicate that there is some annealing of radiation effects with time at temperature at the lower stresses.

Tensile Test Results for Experiments No. 1, 2, and 3 - Rene' 41 test specimens from two heats (TV-746 and TV-754) were used in Experiments No. 1, 2, and 3. The specimens in Heat TV-746 were fabricated from 3/8-inch plate stock,\* while those of Heat TV-754 were fabricated from 7/16-inch round stock.\* The tensile data for the control and irradiated test specimens are listed in Tables 3.4 to 3.7. The marked variations in the tensile properties between specimens of the two heats make it desirable to examine the results separately.

TABLE 3.3

STRESS RUPTURE PROPERTIES OF RENE' 41 AT 870°C

EXPERIMENT NO. 4

11/1/2				EXPERIMEN	T NO. 4	
	Ş	Specimena	Stress, psi	Rupture Time, hr	Elongation, % in 1.5 in.	Reduction Of Area,
		Controls - Sir	nulated cy	cle, 100 hours at	870°C '	
5 K	$v^{\theta_{i}}$	35R 💝		9. 3 1200. 5d	10. 5 12. 0	30. 2 25. 9
		Underaged <sup>C</sup>			7.5	10.0
. 05	on,	24R < ′ 25R ⊙ ✓	20,000	444.6	7. 5 15. 3	19. 8 38. 3
				and the second second		and the second s
.,,		Irradiated at	870°C in N	ит-51:	155 COS 6	10 100 mg 1 20 8
		Irradiated at Overagedb	870°C in <b>N</b>	MT-51:	# \$5° € 00 € 6	in improve lab
		_			0. 7	4.7
o <sup>\$</sup>	بالأون	Overagedb 31R or 32R	25, 000 10, 000	7. 0 4191. 0 <sup>e</sup>	0. 7	
o <sup>St.</sup>	o <sup>a</sup>	Overagedb 31R	25, 000 10, 000	7. 0 4191. 0 <sup>e</sup>		
o <sup>\$</sup>	() <sup>(4)</sup> .	Overagedb 31R or 32R	25, 000 10, 000	7. 0 4191. 0 <sup>e</sup>	0. 7 2. 2	4.7
O.B.	O <sup>M</sup>	Overagedb 31R 0 32R 33R 33R	25, 000 10, 000 17, 500	7. 0 4191. 0 <sup>e</sup>	0. 7 2. 2	4.7
ob.	0 <sup>th</sup>	Overagedb 31R 0 32R 2 33R 2 Underagedc	25, 000 10, 000 17, 500 25, 000	7. 0 4191. 0 <sup>e</sup> 223. 4	0. 7 2. 2 1. 9	4. 7

aAll specimens were of the smooth-bar type from Heat TV-746.

bHeat treatment prior to grinding:

<sup>1205°</sup>C - 2 hr - water quench

<sup>1065°</sup>C - 6 hr - water quench

<sup>925°</sup>C - 48 hr - air cool

<sup>760°</sup>C - 100 hr - air cool

CHeat treatment prior to grinding was the same as above plus 1175°C for 1/2 hr and cooled to 25°C in argon.

dEquivalent time at 870°C based on 43.3 hr at 890°C and 1045.5 hr at  $870^{\circ}$ C.

eEquivalent time at  $870^{\circ}\text{C}$  based on 1003 hr at  $890^{\circ}\text{C}$  and 575 hr at  $870^{\circ}\text{C}$ .

fEquivalent time at  $870^{\circ}$ C based on 98.6 hr at  $890^{\circ}$ C and 698 hr at  $870^{\circ}$ C

<sup>\*&</sup>quot;High-Temperature Materials Program Progress Report No. 13, Part A," GE-NMPO, GEMP-13A, July 31, 1962, Table 5.2, p. 34.

TABLE 3.4
RENE' 41 CONTROL SPECIMEN TENSILE TEST DATA
EXPERIMENTS 1, 2, AND 3

Area, % Temperature, °C Time, hr Quench/Cool Temperature, °C Time, hr 1065 145 2 Water 650 120 100 1065 142 Air 650 1100 1065 1700 166 16 Air 650 1100 1065 1700 166 16 Air 650 1100 1065 1700 166 16 Air 650 1100 1065 1700 1100 1100 1100 1100 1100 1100 110	Test Temperature, 0.2	0. 2-Percent	Ultimate Strength,	Elongation,	Reduction Of	-Pre-	Pre-Treatment		Pre-T	Pre-Test Treatment	ent
8.0 1065 1950 2 Water 650 100 100 100 100 100 100 100 100 100 1	Yield, psi psi	psi		%	Area, %	Temperature, <sup>o</sup> C	Time, hr	Quench/Cool	Temperature, <sup>o</sup> C	Time, hr	Cool/Quench
8.0 1065 17 2 Water 650 12 100 Air 13.8 1065 2 Water 650 100 Air 160 Air 1760 16 Air 1760 16 Air 1760 16 Air 1760 1760 1760 1760 1760 1760 1760 1760	,				( ) ( ) ( ) ( ) ( ) ( ) ( ) ( ) ( ) ( )	03					į
5 13.8 1065 2 Water 650 100 Mater 1065 1/2 Air 1175 1/2 Air 1065 1/2 Air 1175 1/2 Air 1065 1/2 Air 1175 1/2 Air 1065 1/2 Air 1176 1/2 Air 1177 1/2 A	2 126,700 188,600 1		-	10.7		1065	2	Water Air	650 12 <sub>0</sub> 2	100	Air
5 13.8 1065 2 Water 650 100 Air 760 165 1/2 Air 770 1065 1/2 Air 870 1100 100 100 100 100 100 100 100 100					(	760	16 16	Air			
1065   1/2   Air     48.8	$_{\delta} \sim 134,500$ 195,000 13.		13.	വ	13.8	1065	2	Water	650	100	Air
48.8 1065 2 Water 870 100 100 100 100 1005 1/2 Air 870 100 100 1005 1/2 Air 870 100 100 1005 1/2 Air 870 100 100 1005 1/2 Air 8.3 1065 2 Water 650 100 100 1175 1/2 Air 650 100 100 1175 1/2 Air 650 100 100 1175 1/2 Air 650 100 100 100 1175 1/2 Air 870 1005 2 Water 650 100 100 100 1175 1/2 Air 870 1005 2 Water 870 100 100 100 100 1005 2 Water 870 100 100 100 100 1005 2 Water 870 1005 2 Water 870 100 100 100 1005 2 Water 870 1005 2 Wat						1065	1/2	Air			engr. S
48.8 1065 2 Water 870 100 1 62.5 1065 2 Water 870 100 1 62.5 1065 2 Water 870 100 1 62.5 1065 2 Water 870 100 1 63.1 1175 1/2 Air 650 100 1 13.8 1065 2 Water 650 100 1 14.2 Air 650 100 1 15.4 1065 2 Water 870 100 100 100 100 100 100 100 100 100 1						. 097	16	Air	(		
62. 5 $\frac{1065}{1065}$ 1/2 Air 870 100 166 5 1/2 Water 870 100 1005 1/2 Air 760 16 Air 760 16 Air 760 16 Air 8.3 1065 2 Water 650 100 1175 1/2 Air 900 1 4 Air 650 100 100 1175 1/2 Air 870 100 100 100 100 100 100 100 100 100 1	o' 35,300 61,400 20.7	20.	20.		48.8	1065	2	Water	870 15	100	Air
62.5 (27 760 16 Air 870 100 100 1005 1/2 Air 1005 1/2 Air 1005 1/2 Air 1005 1/2 Air 1175 2 1/2 Air 1175 1/2 Air 1005 1/2 Air 1175 1/2 Air 1005 2 Water 650 100 100 1175 1/2 Air 650 100 100 1175 1/2 Air 650 100 100 100 1175 1/2 Air 870 1005 2 Water 870 100 100 100 1005 2 Water 870 100 100 100 1005 2 Water 870 100 100 100 1005 2 Water 870 100 100 1005 2 Water 870 100 100 100 1005 2 Water 870 100 100 100 100 1005 2 Water 870 100 100 100 100 100 100 100 100 100 1							1/2	Air			
62. 5 1065 2 Water 870 100 100 1065 1/2 Air 870 100 100 1065 1/2 Air 650 100 100 1175 $\frac{1}{2}$ 2 Water 650 100 100 1175 $\frac{1}{2}$ 2 Water 650 100 100 1175 1/2 Air 650 100 1175 1/2 Air 650 100 1175 1/2 Air 650 100 100 1175 1/2 Air 870 1000 1000 1175 1/2 Air 870 1000 1000 1175 1/2 Air 870 1000 1000 1005 2 Water 870 1000 1000 1175 1/2 Air 870 1000 1000 1000 1000 1000 1000 1000							16	Air			
8.3 1065 1/2 Air  8.3 1065 1/2 Water 650 100  9.8 1175 2 Water 650 100  9.8 1175 1/2 Air  900 4 Air  900 4 Air  870 100  100 100  1175 1/2 Air  900 4 Air  870 100  100 100  1175 1/2 Air  900 4 Air  870 100  100 100  1175 1/2 Air  900 4 Air  870 100  100  100  100  100  100  100  10	46,000 68,400 24.7		24.7				2	Water	870	100	Air
166 Air  1065  1175  1175  1065  2 Water 650  100  1175  1175  1172  1175  1172  1175  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  1170  117	,					1065	1/2	Air			`,
1065 2 Water 650 100 1175 2 0 4 Air 900 1 2 Water 650 100 1175 1/2 Air 1065 2 Water 650 100 1065 2 Water 650 100 1175 1/2 Air - 900 4 Air 1175 1/2 Air 1065 2 Water 870 100 1065 2 Water 870 100 1065 2 Water 870 100 1065 2 Water 900 4 Air 1175 1/2 Air 900 4 Air 900 4 Air 1175 1/2 Air 900 4 Air 900 4 Air 900 4 Air 1175 1/2 Air 900 4 Air						760	16	Air			)
1175 2 Air Air 900 1/2 Air 650 100 1005 2 Water 870 1005 2 Water 870 100 1005 2 Water 870 1005 2 Water 870 100 1005 2 Water 870 1005 2 Water 870 1005 2 Water 870 1005 2 Water 870 100 1100 1175 1/2 Air 900 4 Air Air 870 100 4 Air 870 100 1100 1175 1/2 Air 870 100 1100 1175 1/2 Air 870 1100 1175 1/2 Air 870 1100 1175 1/2 Air	O 150,700 199,000 9.9		6.6	3	8.3	1065	2	Water	650	100	Air
900 105 4 Air 900 105 1065 2 Water 650 100 1775 1/2 Air 650 100 100 4 Air 1065 2 Water 650 100 1005 2 Water 650 100 100 1005 2 Water 870 100 100 1005 2 Water 870 100 100 1005 2 Water 870 100 1005 2 Water 870 100 1005 2 Water 870 100 1005 2 Water 870 1005 2 Wate						1175 27	1/2	Air			
1175 1, 2 Water 650 100  1175 1/2 Air  900 4 Air  11065 2 Water 650 100  1175 1/2 Air  900 4 Air  1175 1/2 Air  900 4 Air  11065 2 Water 870 100  1065 2 Water 870 100  1175 1/2 Air  900 4 Air  900 4 Air  1175 1/2 Air  900 4 Air						006 مارد	4	Air			
1175 1/2 Air  1065 2 Water 650 100  1175 1/2 Air  107 1175 1/2 Air  1085 2 Water 870 100  4 Air  1085 2 Water 870 100  4 Air  1085 2 Water 870 100  1085 2 Water 100  1175 1/2 Air  900 4 Air  1085 2 Water 870 100  4 Air  900 4 Air  900 4 Air  900 4 Air  900 4 Air  1175 1/2 Air  900 4 Air  900 4 Air  1175 1/2 Air	$c^{\infty}$ 133,700 192,800 11.6		11.6		<b>&amp;</b>		2	Water	650	100	Air
1065 2 Water 650 100 1175 1/2 Air  - 900 4 Air  1065 2 Water 870 100 1175 1/2 Air  900 4 Air  1065 2 Water 870 100 1065 2 Water 870 100 1175 1/2 Air  1065 2 Water 870 100 1175 1/2 Air  900 4 Air  1065 2 Water 870 100 1175 1/2 Air  900 4 Air  1175 Air  900 4 Air  1175 Air  900 4 Air					,		1/2	Air			
1065 2 Water 650 100 1175 1/2 Air  - 900 4 Air  1065 2 Water 870 100 1175 1/2 Air  900 4 Air  900 4 Air  1175 1/2 Air  900 4 Air  1175 1/2 Air  900 4 Air						006	4	Air			
1175 1/2 Air  - 900 4 Air  1065 2 Water 870 100  1175 1/2 Air  900 4 Air  1065 2 Water 870 100  - 1175 1/2 Air  900 4 Air  1175 1/2 Air  900 4 Air  900 4 Air  900 4 Air  1175 1/2 Air  900 4 Air  900 4 Air  1175 1/2 Air  900 4 Air	ු <sup>2</sup> 130, 400 187, 600 14. 4		14. 4		13.8	1065	2	Water	650	100	
1065 2 Water 870 100 1175 1/2 Air 900 4 Air 1065 2 Water 870 100 1065 2 Water 870 100 4 Air 1065 2 Water 870 100 1175 1/2 Air 1175 1/2 Air 900 4 Air						1175	1/2	Air			~
1065 2 Water 870 100 1175 1/2 Air 900 4 Air 1065 2 Water 870 100 1065 2 Water 100 1175 1/2 Air 1065 2 Water 870 100 1175 1/2 Air 1065 2 Water 870 100 4 Air 900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air 900 4 Air 1175 1/2 Air 900 4 Air 900 4 Air 900 4 Air						006 -	4	Air	•		a v)
1175 1/2 Air 900 4 Air 1065 2 Water 870 100 1175 1/2 Air 1065 2 Water 870 100 4 Air 1175 1/2 Air 900 4 Air 1175 2 Water 870 100 4 Air 900 4 Air 900 4 Air 900 4 Air 900 4 Air	45,600 68,300 22.9		22.9		53.7	1065	2	Water	870	100	Air
900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air 11065 2 Water 870 100 1175 1/2 Air 900 4 Air 1175 2 Water 870 100 4 Air 900 4 Air 900 4 Air						1175	1/2	Air			
1065 2 Water 870 100  1175 1/2 Air 1005  1065 2 Water 870 100  1175 1/2 Air 100  4 Air 870 100  4 Air 1065 2 Water 100  1175 1/2 Air 100  1175 1/2 Air 870 100  4 Air 870 100						006	4	Air			
1175 1/2 Air 900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air 1175 1/2 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air	$5^{\circ\prime}$ 135, 100 161, 300 2. 0		2.0				7	Water	870	100	Air
900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air							1/2	Air			
1065       2       Water       870       100         1175       1/2       Air       900       4       Air         1065       2       Water       870       100         1175       1/2       Air       4       Air					.,		4	Air			
1175 1/2 Air 900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air	3.7 114, 200 180, 200 4.1		4.1		5.4	1065	2	Water	. 870	100	Air
900 4 Air 1065 2 Water 870 100 1175 1/2 Air 900 4 Air	`					1175	1/2	Air			
60.4 1065 2 Water 870 100 1175 1/2 Air 900 4 Air						006	4	Air			
1/2	C 57 54, 200 79, 500 26.		26.	7	60.4	1065	2	Water	870	100	Air
4						1175	1/2	Air			5
						006	4	Air			

TABLE 3.4 (Contd.)

RENE' 41 CONTROL SPECIMEN TENSILE TEST DATA EXPERIMENTS 1, 2, AND 3

Propertium		Test	1		Ultimate									ı
870	mena			). 2-Percent Yield, psi	Strength, psi	Elongation, $^{\mathscr{Q}}_{\mathscr{R}}$	Reduction Of Area. %		Pre-Treatment OC Time, hr	Onench/Cool	Pre-To Temperature, <sup>OC</sup>	est Treatme Time. hr	nt Cool/Quenc	1-5
SPO		,	-			2	2, (		( )	<i>(</i>			,	: 1
1175   1/2   Air   1065   2 Water   0.5			0	,	76, 200		33.0	/ 1065	2	Water	870	100	Air	
870 02 64,200 75,400 11.4 24.0 02 1065 2 Water 870 100 Air 1075 1.2 Air 1075 1.2 Air 1075 1.2 Water 870 1.00 Air 1075 1.2 Air 1.								1175	1/2	Air	1065	83	Water	~, ;
Specimens   Spec								006	4	Air	1175	1/2	Air	ه ک
870 0 64, 200 75, 400 11.4 24.0 0 105 2 Water 870 100 Air 1065 2 Water 870 100 Air 1175 1/2 Air											006	4	Air	
1175   1/2   Air   11065   2   Water   870   6	<b>r</b> 0		20	64, 200	75, 400	11.4		~ (	2	Water	870	100	Air	
Specimens   Spec									1/2	Air	1065	7	Water	
Specimens   Spec								006	4	Air	1175	1/2	Air	
870 0.7 75,700 83,800 20.5 39.3 1065 2 Water 870 100 Air 1065 2 Water 150    175 1/2 Air 1065 2 Water 175 1/2 Air 1175 1/2											.006	4	Air	
175   1/2   Air   1065   2   Water   107   1/2   Air   1/2   Ai	148RS		33	75, 700	83,800		39.3	1065	87	Water	870	100	Air	
Specimens   175   175   172   Air   175								1175	1/2	Air	1065	7	Water	
Specimens   Spe						,		006	4	Air	1175	1/2	Air	
Specimens   Spe								,			006	4	Air	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	hed Spe	ecimens						A COMPANY OF THE PARK OF THE P			Company of the second of the s	1		7
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	م.		0	1	233, 400	ı	ı	, 1065	2	Water	650	100	Air	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								1065	1/2	Air				4
650 $\circ$ ~ - 234,400 \( \begin{array}{c ccccccccccccccccccccccccccccccccccc									16	Air			<b>∪</b> .>	~
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	٩		7		234, 400	•	•	<u></u>	23	Water	650	100	Air	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		)	)					/ 1065	1/2	Air		,	~)	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$								7 760	16	Air			,	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		:	5	1	235, 600		1	1065	2	Water	650	100	Air	,
26 5 - 233,500 1065 2 Water 650 100 Air C 650 0 0 Air C 650		74.						1175	1/2	Air				
26 JV -       233,500 -       -       -       1065       2       Water       650       100       Air         650 03 -       220,200 -       - <td< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>006</td><td>4</td><td>Air</td><td></td><td></td><td></td><td></td></td<>								006	4	Air				
650 0 3 - 220,200 (1065 2 Water 650 100 $\frac{4}{900}$ Air 650 0 100 $\frac{1}{900}$ Air 650 0 100 $\frac{1175}{900}$ 4 Air 870 0 119,000 1065 2 Water 870 100 $\frac{1175}{900}$ 4 Air 870 100			ž	1	233, 500	ı	ı	1065	7	Water	650	100	Air 🗸	
650 0 3 - 220, 200 (1065 2 Water 650 100  1175 1/2 Air  900 4 Air  970 0 4 Air  119, 000 11065 2 Water 870 100  1175 1/2 Air  1175 1/2 Air  100 1006 4 Air							•		1/2	Air				
650 0 3 - 220, 200 (1065 2 Water 650 100 1175 1/2 Air 900 4 Air 870 0 4 Air 870 0 4 Air 1175 1/2 Air 1175 1/2 Air							9	•	4	Air			•	
1/2 Air			č	,	220, 200	•	1	1065	2	Water	650	100	Air	
870 0 4 Air 870 0 4 Air 1065 2 Water 870 100 1175 1/2 Air 2900 4 Air								11175	1/2	Air				
870 0 4 - 119,000 11065 2 Water 870 100 - 1175 1/2 Air 2 100 - 2900 4 Air								006	4	Air				
1/2		870 O	+	ı	119,000	ı	1	≥ 1065	2	Water	870	100	Air	
4								1175	1/2	Air			/	
								006	4	Air				

aAll specimens are from Heat TV-754, bExperiment No. 3.

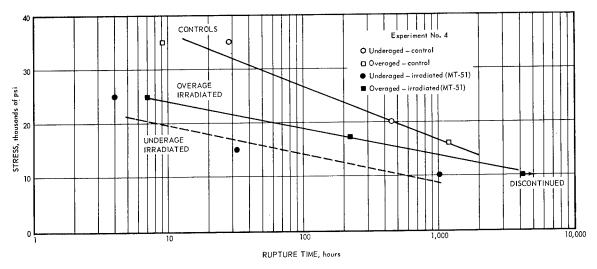


Fig. 3.6-Stress-rupture properties of Rene'41 at 870°C-Heat TV-746, Capsule MT-51

O.J.				TABLE 3	3.5			
1. of.	6000	RENE		OL SPECIME XPERIMENTS	N TENSILE TE S 1 AND 2	ST DATA		
Spacimena	Test V Temperature,	0. 2-Percent	Ultimate Strength,	Elongation,	Reduction Of		-Test Treatme	
√ Specimen <sup>a</sup>	°C	Yield, psi	psi	%	Area, %	Temperature,	°C Time, hr	Cool/Quenci
Smooth Spe	cimens							
(11RS	26 01	99,700	147,000	12. 5	12.4	540	100	Air
33RS	26 °	105,000	144,000	10. 6	12. 3	540	100	Air
81 34RS	540 ° 3	94,500	127, 000	10.0	14. 0	540	100	Air
(35RS	540 o''	95,000	133, 000	12. 0	17.7	540	100	Air
/ 51RS	26 51	117,000	137,000	3. 3	4.8	650	100	Air
\	540 03	110,000	134, 000	5. 6	12. 5	650	100	Air
52RS 53RS	650 e3	103,000	145,000	10. 3	13.9	650	100	Air
141RS	650 s t	101, 900	142, 900	14. 2	11. 4	650	100	Air
/ 41RS	540	89,600	133,000	14. 8	13. 9	650	100	Air
( TINS		00,000				1065	2	Water
						1175	1/2	Air
4)						900	4	Air
رم (54RS	650 c ·	89,600	137, 000	14.6	15.4	650	100	Air
() > )			,			1065	2	Water
/						1175	1/2	Air
1						900	4	Air
√ 36RS	650 🔿 🤼	84,800	136,000	15.0	18.3	650	100	Air
		ŕ	*			1065	2	Water
			•			1175	1/2	Air
						900	4	Air
Notched Spe	ecimens							
04 (26RN	26 o'	_	168,000	_	_	540	. 100	Air
04 (25RN 27RN	540 05	-	180, 000	-	-	540	100	Air
o≦ 11RN	<b>26</b> °	-	171,000	-	-	As	s-machined	
28RN	540 🌣 '	-	202,000	-	- *	650	100	Air
0 - 29RN	650	-	187,000	-	_	650	100	Air

<sup>&</sup>lt;sup>a</sup>All specimens from Heat TV-746, which were heat treated prior to grinding: 1065°C for 2 hr and water quenched plus 1175°C for 2 hr and air cooled plus 900°C for 4 hr and air cooled.

TABLE 3.6

RENE' 41 IRRADIATED SPECIMEN TENSILE TEST DATA EXPERIMENTS 1, 2, AND 3

.,	 					4	EAFERIMENIS 1,	1, 2, AND 9					
		Pre-Irra	Pre-Irradiation Treatment	reatment		Irradiati	Irradiation Conditions		Test		Ultimate		
	~~	Temperature,	Time,			Dosage, nvt	Temperature,	Time, T	Temperature,	0. 2-Percent	Strength,	Elongation,	Reduction Of
Specimen <sup>a</sup>		၁၀		Quench/Cool	Capsule	$(\mathbf{E_n} \ge 1 \; \mathbf{Mev})$	⊃ <sub>0</sub>	hr	- Do	Yield, psi	psi	%	Area, %
Smooth Specimens	)pecin	nens											
$28RS^{C}$	٠	1065	8	Water	MT43	$0.6 \times 10^{20}$	675	67.2	.c 099	145,060	183, 350	5, 60	9.4
		1065	1/2	Air									
	40	160	16	Air		,					;		
$29RS^{C}$		1065	2	Water	MT43	$0.6 \times 1020$	675	67.2	650	143, 210	178,970	4. 73	9.4
	•	1065	1/2	Air									
		160	16	Air /									
26RS		1065	2	Water	MT43	$1.1 \times 1020$	675	67.2	26	141,550	181, 130	13.00	12.5
		1175	1/2	Air									
	`-'	006	4	Air									
27RS	Ç	1065	2	Water	MT43	$1.1 \times 10^{20}$	675	67.2	ووو ووو	132, 450	174,830	8.80	6.3
		1175	1/2	Air									
		006	4	Air		\					,	:	
31RS		1065	2	Water	MT43	0.9 x 1020	675	67.2	650	129,960	165,560	7.47	11.0
	o,	1175	1/2	Air									
	· }	006	4	Air	,		٠						
62RS		1065	2	Water	MT51	0, 6 x 10 <sup>20</sup>	870 /	72.0	26 61	120, 210	141, 710	1.40	6.3
	5	1175	1/2	Air									
		006	4	Air									
63RS		1065	2	Water	MT51	$0.6 \times 10^{20}$	860	72.0	870 01	47,270	52, 970	2. 60	15.4
		1175	1/2	Air									
		006	4	Air		Š		ļ					
64RS	r. K	1065	2	Water	MT51	$0.6 \times 10^{20}$	860	72.0	870 or	54, 790	60, 950	2. 67	4.9
		1175	1/2	Air									
		006	4	Air	, , ,	06					100	10.7	•
71RS		1065	2	Water	MT51	0.6 x 1020	870	72.0	009	125, 410	165, 060	4. 27	٥.
		1175	1/2	Air									
	<	006	4	Air		;		,				,	
73RS	0	1065	2	Water	MT51	$0.6 \times 1020$	, 048	72.0	870 o 55	49, 400	53, 360	3.07	<b>4.</b> 8
		1175	1/2	Air									
		006	4	Air									\
55RS		1065	2	Water	MT51	$0.6 \times 10^{20}$	908	72.0	870 0	63, 360b	65, 060b	3.40b	8. 0b
		1175	1/2	Air									
	5		4	Air		;							1
56RS	5		2	Water	MT51	$0.6 \times 1020$	800	72.0	870 °~	~ 57,363b	58, 2100	3.600	09°6
		1175	1/2	Air									

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TABLE 3.6 (Contd.)

RENE' 41 IRRADIATED SPECIMEN TENSILE TEST DATA

	-						aur (- fr a riverime	1						
	ļ	Pre-Irradiation Treatment	liation Ti	reatment		Irradiati	Irradiation Conditions		Test			Ultimate		
Specimena		Temperature,	Time, hr	Quench/Cool	Capsule	Dosage, nvt $(E_n \stackrel{>}{=} 1 \text{ Mev})$	Temperature, oC	Time, hr	Temperature, oC		0.2-Percent Yield, psi	Strength, E	Elongation, $\%$	Reduction Of Area, %
76RS <sup>c</sup>		1065	2	Water	MT51	0.6 x 1020	870	72.0	870	6	35, 860	40,750	3, 13	7.8
	`	1065	1/2	Air										
	/`i	160	16	Air										
$79RS^{C}$	Ç,	1065	2	Water	MT51	$0.6 \times 10^{20}$	870	72.0	870	}	29,340	31,780	2, 80	10.9
		1065	1/2	Air										
		160	16	Air										
Notched Specimens	Specim	ens	!				:							
17RN <sup>c</sup>		1065	2	Water	MT43	$0.4 \times 10^{20}$	675	67.2	650	i		203, 730	•	,
		1065	1/2	Air										
	0	160	16	Air										
$18RN^{c}$	١.	1065	2	Water	MT43	$0.4 \times 10^{20}$	675	67.2	650	۲,		198,840	1	1
		1065	1/2	Air										
		760	16	Air			The Control of the Co	Table and the control of		;			,	
13RN		1065	2	Water	MT43	$0.4 \times 10^{20}$	675	67.2	56	Ġ	1	236, 320	1	ı
		1175	1/2	Air										
		006	4	Air										
14RN	;	1065	2	Water	MT43	$0.4 \times 1020$	675	67.2	26	) (i)	r	236, 320	ı	1
	è	1175	1/2	Air										
		006	4	Air						1				
15RN		1065	2	Water	MT43	$0.4 \times 10^{20}$	675	67.2	650	ا الراء	ı	208,610		ı
		1175	1/2	Air								,		
		006	4,	Air										
19RN		1065	2	Water	MT51	$0.6 \times 1020$	870	72.0	870			77, 420		•
	ر	1175	1/2	Air										
	-	006	4	Air										
20RN		1065	2	Water	MT51	$0.6 \times 10^{20}$	088	72.0	870			55,000		1
	Ç	1175	1/2	Air										
	• ;	006	4	Air										
21RN		1065	2	Water	MT51	0.6 x 1020	870	72.0	870			66, 500		1
	 ?.	1175	1/2	Air										
	`	006	4	Air		İ								

aAll specimens from Heat TV-754.

bAfter irradiation, specimens 55RS and 56RS were heat treated at 1065°C for 2 hr and water quenched, plus 1175°C for 1/2 hr and air cooled, plus 900°C for 4 hr and air cooled. cSpecimens applicable to Experiment No. 3

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rable 3.7

RENE' 41 IRRADIATED SPECIMEN TENSILE TEST DATA EXPERIMENTS NO. 1 AND 2

		Irradiation	Conditions		Test	0.2-Percent	Ultimate		Reduction
<b>S</b> pecimen <sup>a</sup>	Capsule No.	Dosage, $(E_n \ge 1 \text{ Mev})$	Temperature, OC	Time, hr	Temperature, <sup>O</sup> C	Yield, psi	Strength, psi	Elongation, %	Of Area,
Smooth Spe	cimens								
{ 2R } 3R	MT27 MT27	$0.7 \times 10^{20}$ $0.7 \times 10^{20}$	550 550	96. 0 96. 0		ୀ 111,650 ତୁ ବ′ 105,940	132, 340 140, 160	6. 70 17. 93	19. 7 19. 7
6R 10S	MT38 MT38	0. $5 \times 10^{20}$ 0. $5 \times 10^{20}$	675 675	71. 3 71. 3	26 650	94, 530 <sup>b</sup> on 83, 530 <sup>b</sup>	159, 720 <sup>b</sup> 124, 840 <sup>b</sup>	21.93 <sup>b</sup> 9.66 <sup>b</sup>	4.7 <sup>b</sup> 13.9 <sup>b</sup>
(1R ∫4R	MT27 MT27	$0.7 \times 10^{20}$ $0.7 \times 10^{20}$	550 550	96. 0 96. 0		ල 111,500 යෝ 101,500	134,000 134,000	-	- -
12S 8RS 10RS	MT38 MT38 MT38	$0.5 \times 10^{20}$ $0.5 \times 10^{20}$ $0.5 \times 10^{20}$	675 675 675	71.3 71.3 71.3	26 650 650	∅ 121,000 № 105,000 № 107,000	139,000 140,500 135,000	- -	- - -
Notched Sp	ecimens		•						
7RN 8RN	MT38 MT38	0. $5 \times 10^{20}$ 0. $5 \times 10^{20}$	675 675	71.3 71.3	650 650	<u>-</u>	153,000 158,000	-	-

aAll specimens from Heat TV-746 were heat treated, prior to irradiation, at 1065°C for 2 hr and water quenched, plus 1175°C for 2 hr and air cooled, plus 900°C for 4 hr and air cooled.

bAfter irradiation, specimens 6R and 10RS were heat treated at 1065°C for 2 hr and water quenched plus 1175°C for 1/2 hr and air cooled, plus 900°C for 4 hr and air cooled.

The ultimate tensile strengths of specimens from Heat TV-754 (Experiment No. 1), which received various thermal and irradiation treatments, are presented in Figure 3.7 as a function of test temperature. The irradiated specimens had approximately 10 percent lower strength than the corresponding control specimens, which received a simulated thermal history, at both irradiation temperatures of 650°C and 870°C. This 10 percent difference in the irradiated and control specimens appears to be constant at all tensile test temperatures. Both the 650°C control and irradiated specimens decrease in strength from room temperature to 650°C whereas the 870°C control and irradiated specimens increase in strength from room temperature to 650°C, then decrease in strength at 870°C.

The specimens from Heat TV-754 (Experiment No. 2), which were given a heat treatment after irradiation or simulated cycle show an increase in average strength, but are still within the same range as specimens that were not given a final heat treatment.

In Experiment No. 3, the specimens received a lower solution temperature treatment than the specimens given the standard heat treatment. At 650°C, the simulated cycle specimens (TV-754) have the same tensile strength as the specimens given the standard heat treatment. At the same temperature, 650°C, the irradiated specimens have higher strength than the specimens given the standard heat treatment (Experiment No. 1). At 870°C, however, the simulated and the irradiated specimens have lower strength than the corresponding specimens given the standard heat treatment.

The ultimate tensile strength of all notched specimens is greater than smooth specimens except at  $870^{\circ}$ C, where the value appears to be nearly the same. This would indicate the material is notch ductile but is approaching notch sensitivity at  $870^{\circ}$ C.

The yield strengths for the specimens (TV-754) used in Experiment No. 1 are presented in Figure 3.8. The specimens that were irradiated at  $650^{\circ}$ C have approximately 5 percent less yield strength at room temperature than the control specimens; at  $650^{\circ}$ C the yield strengths are nearly equal. The specimens irradiated at approximately  $870^{\circ}$ C have 10 percent less yield strength at room temperature and about 10 percent greater yield strength at the  $650^{\circ}$ C test temperature, than the control specimens. The  $870^{\circ}$ C yield strength data

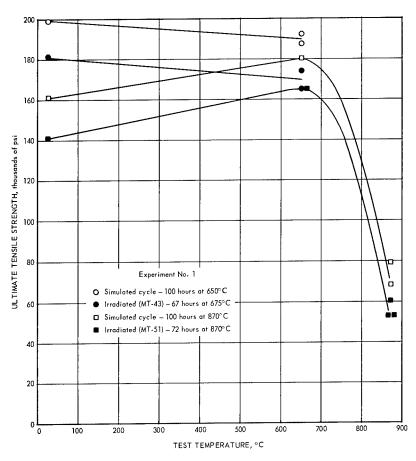


Fig. 3.7 - Tensile strength of control and irradiated smooth Rene'41 specimens - Heat TV-754

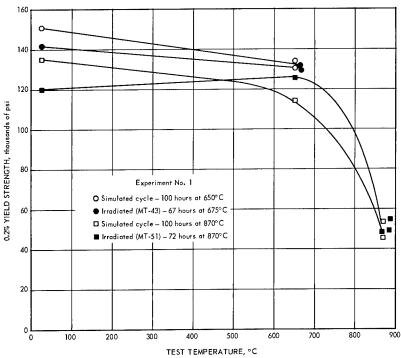


Fig. 3.8-Yield strength of control and irradiated smooth Rene'41 specimens-Heat TV-754

indicate very little effect is caused by irradiation when compared to the control data. The yield strength of irradiated specimens from Experiment No. 3, given the non-standard heat treatment, is much higher than the other specimens at irradiation and test temperatures of 650°C but much lower at irradiation and test temperatures of 870°C.

The percent of elongation for specimens (TV-754) versus test temperatures are shown in Figure 3.9. The marked decrease in ductility for specimens irradiated and tested at 870°C is apparent in the decrease in elongation (and also the reduction in area as listed in Tables 3.4 and 3.6) when compared to control specimens given a simulated cycle at 870°C. There also appears to be a similar trend for the specimens irradiated at 650°C.

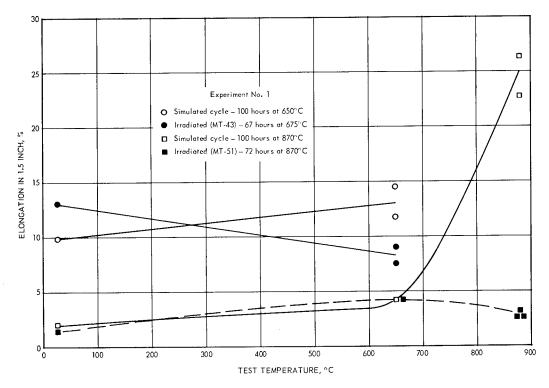


Fig. 3.9 - Elongation of control and irradiated smooth Rene'41 specimens - Heat TV-754

The ultimate tensile strengths for the second heat (TV-746) of Rene' 41, indicate that at room temperature the specimens irradiated at  $540^{\circ}$ C have a lower tensile strength than the control specimens given a simulated temperature cycle. At a test temperature of  $540^{\circ}$ C, the irradiated specimens have a higher strength. This is not the case for the specimens irradiated at approximately  $650^{\circ}$ C; the data indicate that the irradiated specimens are stronger up to  $540^{\circ}$ C than the corresponding control specimens, but weaker at  $650^{\circ}$ C.

Specimens from Heat TV-746 (Experiment No. 2) irradiated at about 650°C, followed by post-irradiation heat treatment, had a higher tensile strength at room temperature but much lower tensile strength at 650°C; the control specimens given the simulated cycle followed by the standard heat treatment had about the same strength at 540°C but a little less strength at 650°C.

The irradiated and control notched specimens from Heat TV-746 also exhibited notch ductility in the temperature range from  $26^{\circ}$ C to  $650^{\circ}$ C when compared to the corresponding control data.

The irradiated specimens (TV-746) have a higher yield strength at both 540°C and 650°C except for the specimens that received the post-irradiation heat treatment (Experiment No. 2) which have a lower yield strength than the corresponding control specimens.

<u>Creep-Rupture Results</u> - Preliminary studies on the creep-rupture properties of Hastelloy X specimens were conducted at 650°C and 870°C. The deformation measurements were made by attaching a dial gage indicator to the upper load rod and also by using an Acotran LVDT transducer which recorded on a strip chart. Both indicators were in good agreement throughout the tests. Three irradiated specimens and three control specimens were tested. The first comparison was made at 650°C and 35,000 psi between a test specimen (32H) irradiated in MT-27 which received a post-irradiation heat treatment and a control specimen (139HS) which received only the standard heat treatment. The irradiated specimen ruptured in 244 hours and the control specimen in 278 hours. These rupture times are comparable to other corresponding test results at the same stress. The creep properties appear to be similar for the two specimens as shown in Figure 3. 10.

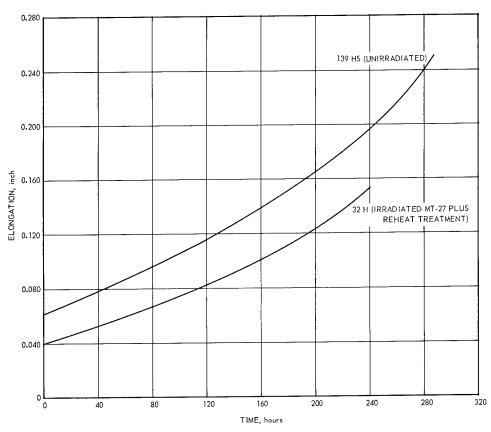


Fig. 3.10-Creep of Hastelloy X at  $650^{\circ}$ C and 35,000 psi

Another comparison conducted at 650°C and 40,000 psi, was between three specimens, one specimen (34H), irradiated in MT-27 and thermally annealed at 650°C for 285 hours, and two control specimens, one as-machined (140HS) and the other receiving a simulated thermal cycle. The irradiated specimen ruptured at 126 hours, the as-machined control specimen at 130 hours, and the simulated specimen at 142 hours. Irradiated specimens that were tested at the same stress but did not receive the initial annealing treatment yielded an average rupture life of 32 hours. The creep curves for specimens 34H and 140HS are plotted in Figure 3.11. The irradiated specimen appears to have a constant

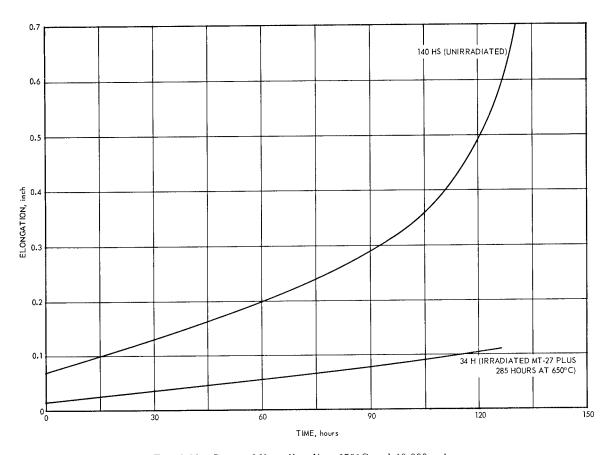


Fig. 3.11 - Creep of Hastelloy X at 650°C and 40,000 psi

creep rate up to the time of rupture. The control specimen, however, shows a large amount of third stage creep.

The deformation curves of Hastelloy X at 870°C are presented in Figure 3.12. In this investigation the irradiated specimen (40HS, MT-51) was thermally annealed at 870°C for 164 hours prior to testing at a stress of 10,000 psi. This specimen ruptured at 16.4 hours, whereas the unirradiated specimen ruptured at 50 hours. Two other irradiated specimens\* tested at the same stress, ruptured at 14.6 and 26 hours. The steady state creep rate for the irradiated and control specimen appear similar. The control specimen exhibited about four times the total elongation of the irradiated specimen which is consistent with previously published data.†

#### STRUCTURE STUDIES

Metallographic examination is continuing on the A-286 samples<sup>‡</sup> that were irradiated specifically for structure studies. The samples were mounted and are currently being prepared for replication.

Electrolytic digestion studies of A-286 stress-rupture test specimens are continuing. To check further for an unidentified phase which was not observed in specimen 10AS during its preliminary analysis but was observed in other A-286 specimens, a rediges-

<sup>\*&</sup>quot;High-Temperature Materials Program Progress Report No. 17, Part A," GE-NMPO, GEMP-17A, November 15, 1962, Figure 4.4, p. 34.

<sup>†</sup>Ibid., Table 4.5, p. 34.

<sup>&</sup>lt;sup>‡</sup>Ibid., Table 4.13, p. 46.

<sup>§</sup>Ibid., p. 48.

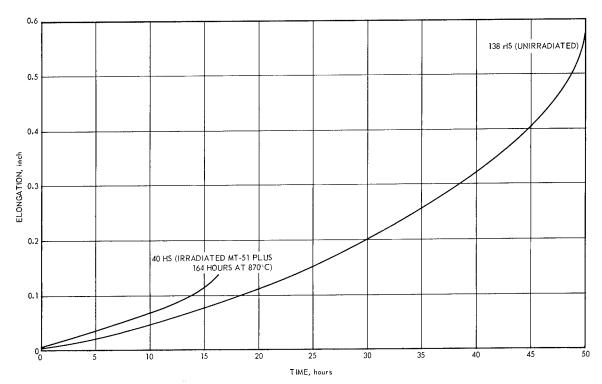


Fig. 3.12 - Creep of Hastelloy X at 870°C and 10,000 psi

tion of this specimen was performed. X-ray diffraction results of the second sampling (10AS-2) again showed only the TiC, Eta, and G phases, with the unidentified phase still not present. The main difference in the A-286 specimens (16K and 9AC) in which the unidentified phase was present was the irradiation temperature; specimens 16K and 9AC were irradiated at 540°C and specimen 10AS at 650°C.

Analysis of the unidentified phase from sample 16K indicates that the material may be face-centered cubic with a lattice parameter of about 10.3Å. These characteristics are similar to those for the carbide phase, M23C6, which is normally not present in A-286. M23C6 is a face-centered cubic structure and has a lattice parameter\* of 10.6Å. Further evaluation will be required before the identity of this phase can be definitely established.

The results of X-ray fluorescence analysis obtained to date are summarized in Table 3.8. Samples 45AS and 16K are from Heat 2527 and samples B15649 and 10AS are from Heat 15649. Some of the data were corrected to a helium path since they were originally measured in an air medium.

Although there is some scatter in the analysis data, it appears that there is no appreciable difference between the concentration of the elements detected for the irradiated and the unirradiated material. Further analysis, however, are required to determine the concentration of carbon or other interstitial elements that were not detected by the X-ray fluoroscope. The results also indicate that the electrolytic digestion variables such as electric current density and acid concentration are not sensitive parameters with respect to the final residue obtained.

<sup>\*</sup>Defense Metals Information Center Memorandum 160, Identification of Micro-constituents in Superalloys, November 15, 1962, p. 11.

TABLE 3.8

APPROXIMATE COMPOSITION OF RESIDUES FROM DIGESTED A-286 SAMPLES

				Ele	ments	a			
<b>S</b> ample	Мо	Nb	Cu	Ni	Co	Fe	Mn	Cr	Ti
45AS-4 <sup>b</sup>	0.7	_	1.6	2. 3	-	4. 2	_	1	8.8
45AS-4	0. 7	_	3.7	2.0	-	4.2	0.2	1	5.4
45AS-1 <sup>b</sup>	1.5	_	_	4.7	_	7.5	-	1	12.6
16K <sup>b,c</sup>	1.1	-	-	4.1	-	6.8	-	1	14.6
B15649-1A	0.9	0.3	5. 3	6. 2	0.1	6.1	0. 2	1	7. 0
B15649-1B	1.0	0. 2	2.2	6.6	0.1	6.4	0.2	1	6.9
B15649-2A	1.4	0. 3	2.1	9.3	0. 2	7.6	0.2	1	10.1
10AS-2 <sup>c</sup>	1.3	0.2	1.3	10.1	0.2	6.5	0.1	1	7. 9

aValues obtained by X-ray fluorescence and arbitrarily normalized to unity for Cr.

bAir path data corrected to equivalent helium path analysis.

cIrradiated sample material.

#### FUNDAMENTAL STUDIES

#### Iron Displacement Studies

A Monte Carlo\* program written for the IBM 7090 computer was used to compute the number and energy spectra of primary knock-on atoms (PKA) produced in neutron-irradiated iron samples. Using this information, the number of displaced atoms was computed on the basis of the Kinchin-Pease displacement model. Neutron energy spectra for a test hole in the ETR reactor† was used to obtain typical results for irradiations in water-moderated reactors. Monoenergetic sources with energies in the range 10 to 10,000 Kev were used to study the energy dependence of the displaced atom population. Experimentally obtained angular distributions for neutron scattering were utilized to account for anisotropic neutron scattering in the center of the mass coordinate system. The effect of neutron slowing down in the irradiated sample and neutron leakage were automatically accounted for by the scattering angle and geometrical tracing routines, respectively, of the Monte Carlo program.

If one assumes the Kinchin-Pease model for atom displacements, the displaced atom density, d, in a monatomic sample is given by:

$$d = (Y/2\beta E_d) [I(O, E^*) + E^*(1-F^*)]$$
 (1)

where

Y = number of PKA produced per incident neutron

Ed = displacement threshold energy

 $\beta$  = volume-to-surface ratio of the irradiated sample

E\* = ionization limit energy for PKA

F\* = fraction of all PKA ejected with energies below E\*.

The number of atom displacements, D, is equal to the product of  $\beta$  and d. I(O, E\*) is the integral:

$$I(O, E^*) = \int_{O}^{E^*} Ef(E)dE$$
 (2)

<sup>\*</sup>J. R. Beeler and J. L. McGurn, "A Method for Computing Primary Knock-On Spectra in Polyatomic Material Irradiated by Neutrons," GE-NMPO, GEMP-101, October 1961.

<sup>†&</sup>quot;High-Temperature Materials Program Progress Report No. 15, Part A," GE-NMPO, GEMP-15A, September 14, 1962, Figure 4.12, p. 53.

where f(e) is the normalized PKA differential energy spectrum. From the definitions of Y and  $\beta$  it follows that d is the number of displaced atoms per unit volume given one incident neutron per unit area of the column surface. Hence, for an isotropic neutron flux,  $\phi$ , and an irradiation time, t, the total displaced atom production (TDP) would be:

$$TDP = d\phi t (3)$$

Following Wollenberger,\* for example, it was assumed that PKA with energies less than E\* lost energy solely by elastic collisions with other atoms, while PKA with energies above E\* lost energy solely by ionization collisions and hence could not produce displaced atoms until they slowed down, below E\*, into the elastic collision range.

In general, Y,  $I(O, E^*)$ , and  $F^*$  are extensive quantities, i.e., they depend upon the size of the irradiated sample and the energy and angular distribution of the incident neutrons. In finite samples, both  $I(O, E^*)$  and  $(1-F^*)$  tend to increase as the neutron energy increases but Y tends to decrease with increasing neutron energy. In iron samples, this competition between opposing influences usually results in a plateua in d for neutron energies in the neighborhood of 1 Mev.

Iron Column - A plot of the atom displacements, D, in a small iron column (square base parallelepiped) is presented in Figure 3. 13, as a function of neutron energy in the range 10 to 10,000 Kev. The length of the column was 6. 35 cm and the base dimension was 0. 2814 cm. An isotropic angular distribution for the neutron angle of incidence was used.

For this particular iron column  $\beta=0.069$ . All iron samples are characterized by  $E_d\simeq 25$  ev,  $E^*\simeq 56$  Kev, and  $F^*=1$  for neutron energies below 800 Kev. The supplementary plots of Y, I(O, E\*), and (1-F\*) show how the variation of D with neutron energy evolves.

<u>Semi-Infinite Medium</u> - Neutron leakage tends to harden the PKA energy spectrum by truncating the neutron collision sequence in the high energy region. In general, leakage occurs by either neutron reflection or transmission. In a semi-infinite medium, which is irradiated by neutrons entering at its free surface, leakage is due solely to reflection, and, for this reason, the characteristics of PKA in a semi-infinite medium are of interest.

The number of displaced atoms, D, in a semi-infinite iron medium is plotted in Figure 3.14, as a function of neutron energy. Supplementary plots of  $I(O,E^*)$ , Y, and  $(1-F^*)$  are also included to show how D evolves as the neutron energy increases.

Below 1 Mev,  $I(O, E^*)$  is the principal factor in the displacement equation. Above 1 Mev, the sharp increase in  $(1-F^*)$  and an increase in Y combine to maintain an increase in D in spite of the slow rise in  $I(O, E^*)$  over this energy range.

Effect of Angular Distribution - If various combinations of the incident neutron angular distribution and the neutron scattering distribution are tried, a significant difference in PKA production is observed for a given neutron energy spectrum. The yield can be changed by a factor of 1.7, the number of displaced atoms by a factor of 1.6, and the average PKA energy by a factor of 1.3 by varying the source angular distribution and neutron scattering distribution. The number of displacements per PKA is relatively insensitive to these changes, however, having a maximum variation of about 10 percent.

A comparison of the atom displacement density for slabs and columns is presented in Figure 3.15. The results for isotropic neutron scattering are compared with those for

<sup>\*</sup>H. Wollenberger, Nukleonik, Vol. 4, 1962, p. 25.

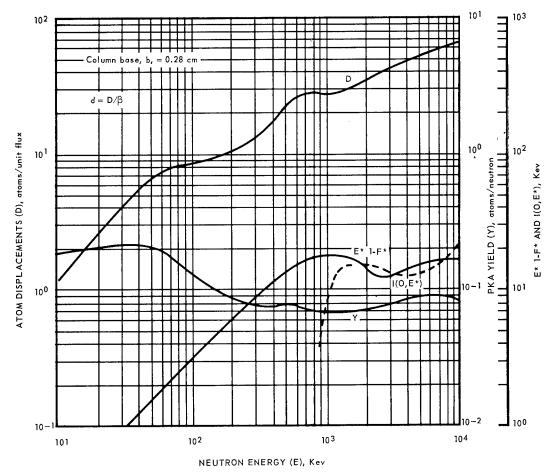


Fig. 3.13 - Displacement parameters in a finite (0.28 by 0.28 by 6.35 cm) iron medium as a function of neutron energy from isotropic neutron source

anisotropic neutron scattering. Normally, incident neutrons with the ETR energy spectrum were used in these calculations. The displacement density is larger in columns with base b than in slabs with thickness t = b; the atom displacement density, d, for isotropic neutron scattering is about 1.5 times that for anisotropic neutron scattering.

#### CONCLUSIONS

Irradiation appears to affect underaged Rene' 41 test specimens more than overaged test specimens when irradiated and tested at a temperature of 650°C. The overaged specimen control and irradiated test data was within 5 percent of the corresponding data for specimens which received the standard heat treatment.

Overaged and underaged Rene' 41 test specimens that were irradiated and tested at 870°C did not show the pronounced difference noted at the lower test temperature.

Irradiation at 650°C and 870°C reduces the ultimate tensile strength of Rene' 41 by approximately 10 percent at all test temperatures. The tests indicate that Rene' 41 is notch ductile at the lower temperatures and approaches notch sensitivity at 870°C.

Elongation and reduction in area data for specimens irradiated at 870°C show ductilities comparable to those of the control specimens at all test temperatures except 870°C where there is a marked decrease.

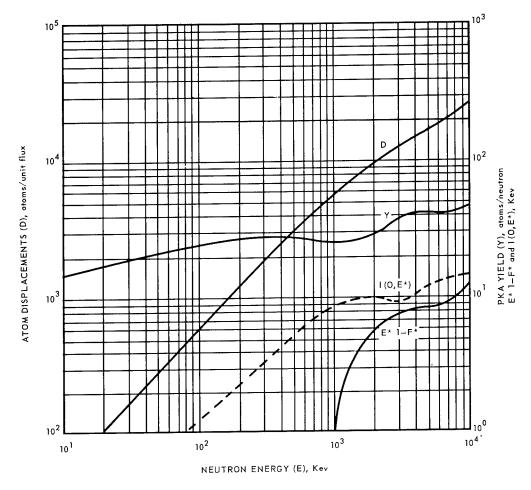


Fig. 3.14 - Displacement parameters in a semi-infinite iron medium as a function of neutron energy from an isotropic neutron source

Atom displacement production by primary knock-on atoms with energies greater than the ionization threshold energy was found by theoretical calculations to be enhanced in specimens of finite geometry relative to the displacements in the semi-infinite systems. The atom displacements will also depend on the angular distribution of neutrons incident on the material and also on the angular dependence of scattering within the medium.

The correlation of radiation effects, based on atom displacements, of in-reactor specimen tests of small dimensions with the corresponding effects in reactor pressure vessels of large dimensions will depend significantly on the material geometry and the angular distribution of the incident neutrons in addition to the neutron spectrum.

#### WORK PLANNED FOR NEXT PERIOD

The preliminary program on refractory metals will be continued with the creep-rupture testing of irradiated tungsten and molybdenum flat specimens. Refractory metal specimen fabrication and specimen irradiation will continue. Furnace operation and testing techniques will be established for handling radioactive test specimens.

Stress-strain diagrams of the irradiated and control tensile data will be evaluated in more detail to further interpret the effects of irradiation on the tensile properties of A-286, Rene' 41, and Hastelloy X.

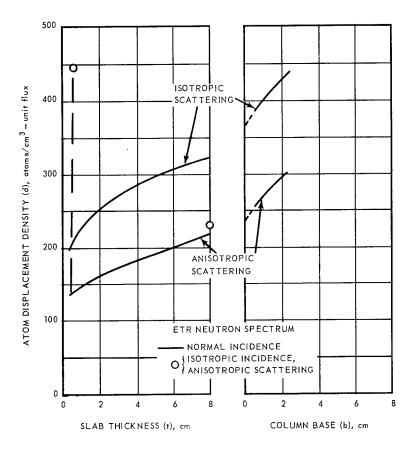


Fig. 3.15 - Atom displacement density for slab and column geometry

Work will continue on the theoretical predictions of atom displacements and the rate of anneal for various times and temperatures. Studies on a preliminary physical model to explain the changes in stress-rupture properties caused by neutron irradiation will continue.

Structure and fracture studies will continue with major emphasis on the A-286 alloy. Digestion techniques for separating the precipitates from the matrix will be continued with special attention to determining whether an unknown phase is formed due to the irradiation.

# 4. LONG TERM IRRADIATION TEST OF 80Ni - 20Cr FUEL ELEMENT SPECIMEN

(57301)

The objective of this task is to establish, on a preliminary basis, the stability of  $80\mathrm{Ni}$  -  $20\mathrm{Cr}$  fuel ribbon at elevated temperatures in air when subjected to neutron irradiation for long periods of time corresponding to burn-up levels of 25 percent of the  $\mathrm{U}^{235}$  atoms present.

No further work was done during this reporting period. A final summary will be included in the 1963 High-Temperature Materials Annual Report.

### 5. APPENDIX

The "High-Temperature Materials Program Progress Reports" previously issued in this series are listed below. The first two reports were each issued as one document, containing both the classified and unclassified portion. The subsequent reports were issued as two documents; part A, the unclassified portion and part B, the classified portion.

Report No.	Report Period	Publication Date
NMP-HTMP-1	May 1961 - June 30, 1961	July 15, 1961
GEMP-2	July 1, 1961 - July 31, 1961	August 15, 1961
GEMP-3, A and B	July 1, 1961 - August 31, 1961	September 15, 1961
GEMP-4, A and B	August 1, 1961 - September 30, 1961	October 15, 1961
GEMP-5, A and B	August 15, 1961 - October 15, 1961	November 15, 1961
GEMP-6, A and B	September 15, 1961 - November 15, 1961	December 15, 1961
GEMP-7, A and B	October 15, 1961 - December 15, 1961	January 15, 1962
GEMP-106, A and B (First Annual Report	) Calendar Year 1961	February 28, 1962
GEMP-9, A and B	January 1, 1962 - February 15, 1962	March 30, 1962
GEMP-10, A and B	January 1, 1962 - March 15, 1962	April 16, 1962
GEMP-11, A and B	February 15, 1962 - April 15, 1962	May 15, 1962
GEMP-12, A and B	March 15, 1962 - May 15, 1962	June 15, 1962
GEMP-13, A and B	April 15, 1962 - June 15, 1962	July 31, 1962
GEMP-14, A and B	May 15, 1962 - July 15, 1962	August 15, 1962
GEMP-15, A and B	June 15, 1962 - August 15, 1962	September 14, 1962
GEMP-16, A and B	July 15, 1962 - September 15, 1962	October 15, 1962
GEMP-17, A and B	August 15, 1962 - October 15, 1962	November 15, 1962
GEMP-18, A and B	September 15, 1962 - November 15, 1962	December 14, 1962